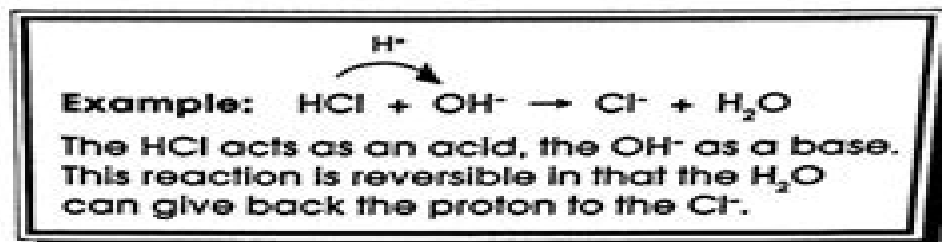


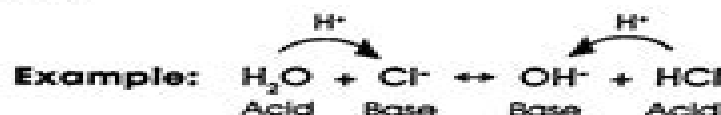
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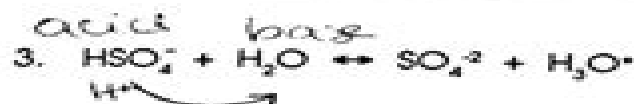
According to Bronsted-Lowry theory, an acid is a proton (H^+) donor, and a base is a proton acceptor.



Label the Bronsted-Lowry acids and bases in the following reactions and show the direction of proton transfer.



acid = donor
base = acceptor



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Clifford W. Hand, Harry Lyon Blewitt



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