

Acids, Bases, Buffers and pH Data Tables

Part A.

	Solution	pH Meter		pH paper		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value	color	pH value
	Control (H_2O)	5.20	6.30×10^{-6}	yellow	6.0	yellow	7.0
a.	0.1 M NH_3	11.29	5.24×10^{-12}	blue	10.0	purple	10
b.	0.1 M NaOH	12.94	1.14×10^{-13}	blue	10.0	purple	10
c.	0.1 M H_3BO_3	6.60	2.5×10^{-7}	yellow	6.0	yellow	6
d.	0.1 M HCl	0.80	1.5×10^{-1}	red	2.0	red	3

Part B.

	Solution	pH Meter		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value
	Control (H_2O)	5.20	6.30×10^{-6}	yellow	7.0
a.	0.1 M Na_2CO_3	11.48	3.31×10^{-12}	purple	10
b.	0.1 M $(NH_4)_2SO_4$	7.40	3.98×10^{-8}	orange	5
c.	0.1 M NaCl	7.96	1.09×10^{-8}	green	6

Part C.

	Solution	pH Meter
		Reading
	Control (H_2O)	5.20
a.	Acetic Acid	3.26
b.	Sodium Acetate	8.11
c.	Mixture of a & b	4.95

Buffers In Household Products Pre Lab Answers

Guillaume Favre



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