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David Baud

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web rtca do 160 section 5 temperature variation testing rtca do 160 section 5 determines performance characteristics of the

equipment during temperature variations between high and low operating temperature extremes note the test is not intended to verify the behavior of the equipment in wet or icing conditions

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web category designation for equipment shall consist of six characters pin test waveform set letter a or b as designated in table $22\ 1\ 1$ or z or x pin test level 1 to 5 as designated in table $22\ 2$ or z or x cable bundle single and multiple stroke test waveform set letter c through k as designated in table $22\ 1\ 2$ or z or x

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web the tests in rtca do 160 provide a laboratory means of demonstrating the performance characteristics of airborne equipment in environmental conditions that may be encountered in operation of the equipment in aircraft it is not the intent of rtca do 160 to be used as a measure of service life of the airborne equipment subjected to these tests

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web apr 9 2020 while iec6008 2 provides guidance for environmental tests of electronic and electro mechanical equipment rtca do 160e section 4 provides standard temperature test conditions and applicable test procedures for airborne equipment environmental test chamber temperature test chamber

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web rtca do 160 section 4 0 temperature and altitude the temperature and altitude tests determine the performance characteristics of equipment at the applicable categories for the temperatures and altitudes specified in table 4 1 and at the pressures defined in table 4 2 the following categories cover the wide range of environments known

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section learning objectives by the end of this section you will be able to do the following describe the structure and forces present within the nucleus explain the three types of radiation write nuclear equations associated with the various types of radioactive decay

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notes 323 nuclei and radioactivity physics module 7 atoms and nuclei z explain the growth and decay of radioactivity in a

sample z calculate the half life and decay constant of a radioactive substance and z explain the uses of radioactivity in various fields 26 1 the atomic nucleus soon after the discovery of nucleus in an atom by rutherford in 1911

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quantum physics yet even though it has only 10 15 of the volume of the atom the nucleus contains more than 99 9 of the mass of the atom the nucleus is very small but very massive that was not predicted try to imagine the surprise and disbelief of scientists in 1911 when ernest rutherford discovered this incredible fact

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over 3 000 different kinds of atomic nuclei have been discovered in elements that span the periodic table yet each one has unique properties some may exist forever others transform within a fraction of a second some can power satellites as they travel to the outer reaches of the solar system others have special properties that allow them to treat different forms of cancer

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radioactivity some nuclides have an atomic nucleus that decays spontaneously without any external influence known as radionuclides they transform themselves into other nuclei ionising radiation is emitted during this process this is known as radioactivity

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it some future development of atomic theory perhaps a modification of the thomson model would most likely yield a causal explanation of radioactivity in terms of the internal structure of the atom

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atomic physics deals with the atom as a system consisting of a nucleus and electrons nuclear physics deals with the nucleus as a system consisting of nucleons protons and neutrons the main difference is in the scale while the term atomic deals with $1\text{\AA}\ 10\ 10\ \text{m}$ where $\text{\AA}\ \text{is}$ an angström according to anders jonas Ångström the term

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