

APCO Institute

Emergency Medical Dispatch Guidecards

9th Edition

Chest Pain/Heart Problems

Vital Points Questions

Is the patient breathing normally?
Is the patient conscious?
Where in the chest is the pain located?
Does the patient feel pain anywhere else? When?
How long has the patient had chest pain?
Sudden Onset:
Does the pain change when the person breathes or moves?
What is the age and gender of the patient?
Is the patient weak, dizzy, or faint?
Is the patient nauseated or vomiting?
Is the patient sweating profusely?
Is the patient experiencing rapid heart rate with chest pain?
Does the patient have a history of rapid heart rate?
Does the patient take nitroglycerin? Any relief?
Has the patient taken an aspirin after the onset of pain?
How much did they take?
Any use of prescription or illegal drugs in the last 48 hours?
Information Only Questions:
Does the patient have a history of heart problems?
Is there a heart surgery or heart attack?
Does the patient have any other medical history?
Does the patient have any type of dental device?

BLS Priority

Unconscious, not breathing normally/difficultly breathing
Decreased level of consciousness
Center of chest, upper abdomen
jaw, shoulder blades, throat, back
minutes, hours, days, weeks (Jd will be ALS)
Painful Breathing
Male <25 years of age or Female <40 years of age
15-40 years of age with any of the following symptoms:
Weak, dully, or faint, short of breath
Nausea
Diaphoresis
Rapid heart rate with chest pain, irregular rapid heart rate
History of rapid heart rate
Nitroglycerin more than three times 2
Aspirin -- How many
Small orange pill or large white pill
Cocaine/crack or other illicit drug use / Prescription drugs such
as Viagra, Cialis, Levitra, etc. in the last 48 hours

Pre-Arrival Instructions

If unconscious and not breathing, go to CPR for the appropriate age group. Trained bystanders may still need instructions. **ASAP IF THEY NEED IT! DONT HESITATE!**
If unconscious, go to **APPLY AIR CONTROL** instructions.
Have the patient sit or lie down, whichever is more comfortable and remain calm.
Does the patient have prescribed nitroglycerin?
How the patient takes any?
IF NO: Have the patient sit down and take as the physician has directed. Make sure they take no more than 3 doses. If dizzy do not take.
Patient been instructed by their physician to take an aspirin?
Is the patient allergic to aspirin or have a bleeding disorder?
IF NO: Have the patient sit down, chew one adult aspirin (325 mg) or 4 baby aspirin (81 mg).
• Loosen tight clothing.
• Nothing to eat or drink.
If Applicable:
• Gather patient's medications and give to responders when they arrive.
• Put any pets away.
• Turn the outside light on another wall at the door.
If anything changes in the patient's condition worsens, please call back.

BLS Priority

Male <25 yrs without critical symptoms
Female <40 yrs without critical symptoms
Rapid heart rate without critical symptoms

BLS Standard

Male <25 yrs or Female <40 yrs with chest wall trauma without critical symptoms

ON VERSION

INTRODUCTION

ALL CALLERS & EMD INTERROGATION

ALLERGIC REACTION

BREATHING DIFFICULTY

CHOKING

GYNCOLOGY / OB / PREGNANCY

HEAT / COLD ENVIRONMENTAL

OVERDOSE / POISONING

SICK / UNKNOWN

UNCONSCIOUS / UNRESPONSIVE / SYNCOPED

ASSAULT / SEXUAL ASSAULT

ELECTROCUTION / LIGHTNING

FALLS

MOTOR VEHICLE CRASH (MVC)

NEUROLOGICAL / HEAD INJURIES

TRAUMA

CPR - ADULT

CPR - INFANT

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Table of Contents Apco Law Enforcement Guide Cards

1. Understanding the eBook Apco Law Enforcement Guide Cards
 - The Rise of Digital Reading Apco Law Enforcement Guide Cards
 - Advantages of eBooks Over Traditional Books
2. Identifying Apco Law Enforcement Guide Cards
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Apco Law Enforcement Guide Cards
 - User-Friendly Interface
4. Exploring eBook Recommendations from Apco Law Enforcement Guide Cards

- Personalized Recommendations
- Apco Law Enforcement Guide Cards User Reviews and Ratings
- Apco Law Enforcement Guide Cards and Bestseller Lists
- 5. Accessing Apco Law Enforcement Guide Cards Free and Paid eBooks
 - Apco Law Enforcement Guide Cards Public Domain eBooks
 - Apco Law Enforcement Guide Cards eBook Subscription Services
 - Apco Law Enforcement Guide Cards Budget-Friendly Options
- 6. Navigating Apco Law Enforcement Guide Cards eBook Formats
 - ePub, PDF, MOBI, and More
 - Apco Law Enforcement Guide Cards Compatibility with Devices
 - Apco Law Enforcement Guide Cards Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Apco Law Enforcement Guide Cards
 - Highlighting and Note-Taking Apco Law Enforcement Guide Cards
 - Interactive Elements Apco Law Enforcement Guide Cards
- 8. Staying Engaged with Apco Law Enforcement Guide Cards
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Apco Law Enforcement Guide Cards
- 9. Balancing eBooks and Physical Books Apco Law Enforcement Guide Cards
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Apco Law Enforcement Guide Cards
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Apco Law Enforcement Guide Cards
 - Setting Reading Goals Apco Law Enforcement Guide Cards
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Apco Law Enforcement Guide Cards

- Fact-Checking eBook Content of Apco Law Enforcement Guide Cards
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web now if you want to reverse the operation of the circuit simply wire the blue and red wires together from the timer and connect it to the photo eye as shown in the figure now the circuit is reversed for example the light will turn off at night and switch on in the sunshine during the day good to know it is recommended to add a relay

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web this type of positive feedback is also called hysteresis and provides a fast transition from the on to off state once turning off starts the off delay timing is mostly a function of c1 and r6 the exact delay is a bit hard to predict because it depends on the gain of the transistors particularly of q1

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web answers lewis dot lewis dot structures for each molecule on the worksheet the lewis dot structure the number of valence electrons the electron arrangement e a and the molecular geometry m g are given respectively to see a larger view of the lewis dot structure point at the molecule right click on the mouse and go to view image

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help for each of the following draw the lewis dot structure give the electron arrangement e a and the molecular geometry m g

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web in 1916 american chemist gilbert n lewis introduced bond lines to electron dot structures these structures also known as lewis structures or electron dot structures are drawings that visually demonstrate how electrons are shared and arranged around atoms the electrons denoted as dots are called lone pairs and belong to an individual

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web lewis dot structures are useful for describing chemical bonds but have some flaws a lewis structure is a picture of a molecule that shows the covalent bonds and pairs of free electrons the octet rule is the basis for lewis structures

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web lewis dot formula also called an electron dot formula shows the valence electrons indicating the bonding between atoms the following guidelines will help draw the electron dot formulas correctly

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web lewis structures extend the concept of the electron dot diagram by adding lines between atoms to represent shared pairs in a chemical bond lewis structures show each atom and its position in the structure of the molecule using its chemical symbol

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web lewis dot structures also called electron dot structures are diagrams that describe the chemical bonding between atoms in a molecule they also display the total number of lone pairs present in each of the atoms that constitute the molecule

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web oct 29 2021 contributors and attributions learning objectives to draw lewis structures for molecules and polyatomic ions with one central atom introduction to lewis structures a lewis structure is a way to show how atoms share electrons when they form a molecule lewis structures show all of the valence electrons in an atom or molecule

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web lewis structures we also use lewis symbols to indicate the formation of covalent bonds which are shown in lewis structures drawings that describe the bonding in molecules and polyatomic ions for example when two chlorine atoms form a chlorine molecule they share one pair of electrons

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web this online quiz is intended to give you extra practice in identifying and drawing lewis dot structures as well as predicting ion formation this quiz aligns with the following ngss standard s hs ps1 1

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web jan 30 2023 a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

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web jan 30 2023 follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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