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**Larry Warren, Alan Ahlstrand, John
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web the carbon hydrogen chains in vulcanized rubber are held together by two sulfur atoms that form covalent bonds between the chains these covalent bonds are commonly called disulfide bridges explore other molecules that have such disulfide bridges present your findings to the class

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web because of the fact that polar covalent bonds have higher melting and boiling points why do dipole forces have an impact on the melting and boiling points of substances this is because the dipole forces are an added force of attraction that slows the molecules down making it take more energy to melt or boil them