# AQA A LEVEL PHYSICS PAPER 1

- What is meant by work function (2): Minimum energy required by an electron to escape from (1) a metal surface (1)
- What is meant by ionisation energy? (2): Minimum energy required to remove an electron from an atom (1) from the ground state (1)
- 3. Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4): Energy of a photon is fixed/constant (1)

  Energy is required for electrons to overcome work function Ø (1)

  Electrons deeper into/below the surface of the metal require more energy to over- come work function and therefore have less KE(1)

  Maximum KE = Energy of photon work function Ø (1)
- 4. Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons?: Diffraction pattern maxima/minima (bright and dark areas) (1) Diffraction is a property of waves- proves wave-behaviour of electrons (1)
- 5. What is meant by the ground state?: The lowest energy state of an atom 6. Explain the difference between excitation and ionisation (2): An electron receives a discrete amount of energy for BOTH (1) Excitation promotes an electron to a higher energy level lonisation is when an electron receives enough energy to leave the atom 7. Explain why only photons of certain frequencies cause excitation (4):
- dis- crete energy levels (1)
  - Electrons need to absorb a DISCRETE amount of energy to move to a higher energy level (1)

# **Aqa Physics Unit 1 January 2013 Paper**

**David Kelly** 

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