

Basic Transport Phenomena in Biomedical Engineering 4th Edition Fournier Solutions Manual

Chapter 1 Solutions

Problem 1.1. Use the conversion factors

$$\frac{8.314\text{J}}{\text{mol K}} \times \frac{0.23901\text{cal}}{\text{J}} = 1.987 \frac{\text{cal}}{\text{mol K}}$$

Problem 1.2. Use Equation 1.2 to convert the temperature to degrees Celsius

$$t^{\circ}\text{C} = \frac{5}{9}(t^{\circ}\text{F} - 32) = \frac{5}{9}(98.6 - 32) = 37^{\circ}\text{C}$$

Now, use Equation 1.3 to convert the temperature from degrees Celsius to Kelvin

$$T^{\circ}\text{K} = t^{\circ}\text{C} + 273.15 = 37 + 273.15 = 310.15^{\circ}\text{K}$$

Problem 1.3. Use the conversion factors

$$2 \times 10^4 \mu\text{dynes} \times \frac{\text{dynes}}{10^6 \mu\text{dynes}} \times \frac{\text{N}}{10^5 \text{dynes}} \times \frac{\text{kN}}{10^3 \text{N}} = 2 \times 10^{-10} \text{kN}$$

Problem 1.4. Use the conversion factors

$$\frac{27 \mu\text{m}}{\text{hr}} \times \frac{10^{-6} \text{m}}{\mu\text{m}} \times \frac{0.0006214 \text{miles}}{\text{m}} \times \frac{\text{hr}}{60 \text{min}} \times \frac{\text{min}}{60 \text{sec}} = 4.66 \times 10^{-12} \frac{\text{miles}}{\text{sec}}$$

Problem 1.5. Use the conversion factors

$$48 \text{MW} \times \frac{10^6 \text{W}}{\text{MW}} \times \frac{\text{kg m}^2}{\text{sec}^2 \text{W}} \times \frac{10^3 \text{g}}{\text{kg}} \times \frac{100 \text{cg}}{\text{g}} \times \frac{3600^3 \text{sec}^3}{\text{hr}^3} \times \frac{100^2 \text{cm}^2}{\text{m}^2} = 2.24 \times 10^{27} \frac{\text{cgcm}^2}{\text{hr}^3}$$

$$48 \text{MW} \times \frac{10^6 \text{W}}{\text{MW}} \times \frac{\text{J}}{\text{sec W}} \times \frac{\text{kJ}}{10^3 \text{J}} \times \frac{60 \text{sec}}{\text{min}} = 2.88 \times 10^6 \frac{\text{kJ}}{\text{min}}$$

Problem 1.6. Use the conversion factors to get both values into units of mm per second and then compare the two values for the higher value.

Basic Transport Phenomena In Biomedical Engineering Third Edition

Tobias Bleicker



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Table of Contents Basic Transport Phenomena In Biomedical Engineering Third Edition

1. Understanding the eBook Basic Transport Phenomena In Biomedical Engineering Third Edition
 - The Rise of Digital Reading Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Basic Transport Phenomena In Biomedical Engineering Third Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Personalized Recommendations

- Basic Transport Phenomena In Biomedical Engineering Third Edition User Reviews and Ratings
- Basic Transport Phenomena In Biomedical Engineering Third Edition and Bestseller Lists
- 5. Accessing Basic Transport Phenomena In Biomedical Engineering Third Edition Free and Paid eBooks
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- 6. Navigating Basic Transport Phenomena In Biomedical Engineering Third Edition eBook Formats
 - ePub, PDF, MOBI, and More
 - Basic Transport Phenomena In Biomedical Engineering Third Edition Compatibility with Devices
 - Basic Transport Phenomena In Biomedical Engineering Third Edition Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Highlighting and Note-Taking Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Interactive Elements Basic Transport Phenomena In Biomedical Engineering Third Edition
- 8. Staying Engaged with Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Basic Transport Phenomena In Biomedical Engineering Third Edition
- 9. Balancing eBooks and Physical Books Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Basic Transport Phenomena In Biomedical Engineering Third Edition
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Setting Reading Goals Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Basic Transport Phenomena In Biomedical Engineering Third Edition
 - Fact-Checking eBook Content of Basic Transport Phenomena In Biomedical Engineering Third Edition

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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