

## Acid Conductivity Experiment

Record your experimental results in this table: [page 1 of 10](#)

| Solution  | Measured Conductivity |
|-----------|-----------------------|
| 100% acid | 100000                |
| 100% acid | 100000                |
| 100% acid | 100000                |
| 100% acid | 100000                |

How conductivity is related to strength of acids: [page 2 of 10](#)

Stronger acids dissociate completely in solution, leading to more ions in solution. More ions in solution leads to greater electrical conductivity.

Why is this a weak acid: [page 3 of 10](#)  
Conductivity is 1000

Water

Why is this a weak base: [page 4 of 10](#)  
Conductivity is 1000

# Apex Study Sheet English4 Answers

**Matthew Lanni**



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