

① $x = 0.01$ Cr, $Ni = 0.04$

Mg = $0.5 \times 0.95 = 0.475$ M (Mg nitrate)

$$0.475 = \frac{n}{0.05} \Rightarrow n = 0.02375 \text{ mol}$$

$$\therefore w = 6.0895 \text{ g}$$

Mg nitrate

Ni = $0.5 \times 0.04 = 0.02$ M

$$0.02 = \frac{n}{0.05} \Rightarrow n = 0.001 \text{ mol}$$

$$w = 0.2908 \text{ g}$$

Ni nitrate

Cr = $0.5 \times 0.01 = 0.005$

$$0.005 = \frac{n}{0.05} \Rightarrow n = 0.00025 \text{ mol}$$

$$w = 0.100035 \text{ g}$$

Cr nitrate

② $x = 0.03$ Cr, $Ni = 0.02$

Mg $w = 6.0895 \text{ g}$

Ni = $0.5 \times 0.02 = 0.01$ M

$$\therefore n = 0.0005 \text{ mol Ni}$$

$$w = 0.1457 \text{ g}$$

Ni nitrate

Cr = $0.5 \times 0.03 = 0.015$ M

$$n = 0.00075 \text{ mol}$$

$$w = 0.300105 \text{ g}$$

Cr nitrate

Analytical Chemistry Calculation Questions And Answers

DP Hallahan



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Basic Analytical Chemistry (Penerbit USM) Faiz Bukhari Mohd Suah, Wan Saime Wan Ngah, Che Sofiah Saidin, 2018 BASIC ANALYTICAL CHEMISTRY Malaysia is a fast developing country Realizing the need to provide experts in chemistry this book is appropriate to be used as a text for fundamental course in analytical chemistry The texts cover topics from the most basic analytical chemistry course including methods on basic analyses to important concepts such as handling of data analysis chemical equilibrium stoichiometry and titration The chemical equilibrium in this book covers acid base equilibrium precipitation complex and redox titration For every topic examples and solutions are provided to give reader a better understanding in the topics covered

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D. Christian, Purnendu K. Dasgupta, Kevin A. Schug, 2013-10-07 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry

Foundations of Analytical Chemistry Miguel Valcárcel Cases, Ángela I. López-Lorente, Ma Ángeles López-Jiménez, 2017-08-29 This book offers a completely new approach to learning and teaching the fundamentals of analytical chemistry It summarizes 250 basic concepts of the field on the basis of slides Each of the nine chapters offers the following features Introduction Summary General scheme Teaching objectives Text containing the explanation of each slide Recommended and commented bibliography Questions to be answered Slides A distinct feature of this novel book is its focus on the fundamental concepts and essential principles of analytical chemistry which sets it apart from other books presenting descriptive overviews of methods and techniques

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