

Structure of H_2O
hydrogen bonding
Dissociation of H_2O



unit one guide

biological molecules
proteins
nucleic acids

H_2O properties: cohesion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, pH buffer, temperature stability

hydrogen bonds require lots of energy to change state of matter

COHESION → when H_2O bonds to it produces hydrogen and hydroxyl ions, H_2O dissociates, pH → more hydrogen
Adhesion → attracted to some surface
Surface Tension → hydrogen bonds
High specific heat → surface of the liquid doesn't heat

PHS bond when bond → water expands due to hydrogen bonding, molecules move slowly

C	H	N	O	P	S
Carbon	Hydrogen	Nitrogen	Oxygen	Phosphorus	Sulfur

more hydrogen bonds
C → build biological molecules, cell formation
N → build proteins, nucleic acids
P → build nucleic acids + lipids
→ make up 98% of organisms

Carbohydrates	Lipids	Nucleic acids	Proteins
↓	↓	↓	↓
metabolites	energy	information	enzymes
↓	↓	↓	↓
C, H, O	C, H, O	C, H, N, O, P	C, H, N, O

Condensation synthesis → if the removal of an H_2O molecule to bond monomers
monomers to polymers, a bond between monomers (condensation)

Nucleic acid Structure:

* first carbon sugar (deoxyribose/ribose), phosphate, nitrogenous base (G, C, A, T, U), exact sequence of the code



Starch vs. Cellulose

Starch → α glucose
Cellulose → β glucose

Proteins: Specific order of amino acids
Primary structure → 1st group
Secondary structure → hydrogen bonds determine structure + function

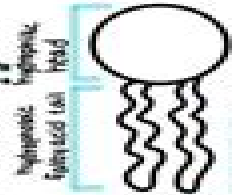


PROTEINS

* amino acids bond via a peptide bond (covalent) through dehydration synthesis
* 1 group change can cause a hydrophobic protein to lose the place of a hydrophilic protein

membranes

* made up of phospholipids
* create selectively permeable barrier
* contain phospholipids, cholesterol, and proteins



LIPIDS

Phospholipid macromolecules
* differences in saturation determine structure and function
* fats, steroids, phospholipids
* Saturated → no double bonds
* Unsaturated → has double bonds
* in fats
* Steroids
* phospholipids



Amino acids: have directionality
* linear chains, connected by covalent bonds
* amino end and carboxyl end

NUCLEIC ACIDS

* add new nucleotides to the 3' end
* two nitrogenous bases: purine + pyrimidine
* purine is single ring, pyrimidine is double
* Antiparallel double helix
* 5' on phosphate end, 3' on hydroxyl end
* both back and forth have nitrogenous bases perpendicular to sugar-phosphate backbone

DEHYDRATION SYNTHESIS



covalent bonds form to create nucleic acids
* the number synthesizes the direction of code

ANTIPARALLEL



Chaperones:

* protein molecules that assist protein folding

PROTEIN STRUCTURE

Primary: Sequence of amino acids, specific order determines structure
Secondary: how folding of chain into alpha helix and beta pleated sheet
Tertiary: 3-D shape, hydrogen bonds, disulfide bond, ionic bond, hydrophobic int.
Quaternary: interactions between multiple polypeptide units

+ FUNCTION

* Enzymes → speeds up reaction rates in the cell
* Cell structure → part of the cellular membrane
* Communication between cells

DENATURATION:

* the protein starts to lose its shape/structure when bonds break
* Temperature change
* pH of surrounding area
* Salt concentration

NH₂ = Amino
COOH = Carboxyl

ALDOSE VS KETOSE

* Contains aldehyde functional group
* Contains ketone functional group



ALPHA VS BETA

* all monomers are in the same organization
* Every glucose monomer is upside down from its neighbor



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web you inspire me and bless me with purpose faithful one you carry me into maturity you hold me as my flesh falters and fails you assure me and help me pass on your blessing

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