

Table 18.2

Solubility Product Constants ( $K_{sp}$ ) at 25°C

| Salt                             | $K_{sp}$              | Salt              | $K_{sp}$              | Salt                            | $K_{sp}$              |
|----------------------------------|-----------------------|-------------------|-----------------------|---------------------------------|-----------------------|
| <b>Halides</b>                   |                       | <b>Sulfates</b>   |                       | <b>Hydroxides</b>               |                       |
| AgCl                             | $1.8 \times 10^{-10}$ | PbSO <sub>4</sub> | $6.3 \times 10^{-7}$  | Al(OH) <sub>3</sub>             | $3.0 \times 10^{-34}$ |
| AgBr                             | $5.0 \times 10^{-13}$ | BaSO <sub>4</sub> | $1.1 \times 10^{-10}$ | Zn(OH) <sub>2</sub>             | $3.0 \times 10^{-16}$ |
| AgI                              | $8.3 \times 10^{-17}$ | CaSO <sub>4</sub> | $2.4 \times 10^{-5}$  | Ca(OH) <sub>2</sub>             | $6.5 \times 10^{-6}$  |
| PbCl <sub>2</sub>                | $1.7 \times 10^{-5}$  | <b>Sulfides</b>   |                       | Mg(OH) <sub>2</sub>             | $7.1 \times 10^{-12}$ |
| PbBr <sub>2</sub>                | $2.1 \times 10^{-6}$  | NiS               | $4.0 \times 10^{-20}$ | Fe(OH) <sub>2</sub>             | $7.9 \times 10^{-16}$ |
| PbI <sub>2</sub>                 | $7.9 \times 10^{-9}$  | CuS               | $8.0 \times 10^{-37}$ | <b>Carbonates</b>               |                       |
| PbF <sub>2</sub>                 | $3.6 \times 10^{-8}$  | Ag <sub>2</sub> S | $8.0 \times 10^{-51}$ | CaCO <sub>3</sub>               | $4.5 \times 10^{-9}$  |
| CaF <sub>2</sub>                 | $3.9 \times 10^{-11}$ | ZnS               | $3.0 \times 10^{-23}$ | SrCO <sub>3</sub>               | $9.3 \times 10^{-10}$ |
| <b>Chromates</b>                 |                       | FeS               | $8.0 \times 10^{-19}$ | ZnCO <sub>3</sub>               | $1.0 \times 10^{-10}$ |
| PbCrO <sub>4</sub>               | $1.8 \times 10^{-14}$ | CdS               | $1.0 \times 10^{-27}$ | Ag <sub>2</sub> CO <sub>3</sub> | $8.1 \times 10^{-12}$ |
| Ag <sub>2</sub> CrO <sub>4</sub> | $1.2 \times 10^{-12}$ | PbS               | $3.0 \times 10^{-28}$ | BaCO <sub>3</sub>               | $5.0 \times 10^{-9}$  |

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**ML Morrison**



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