

## CAMPBELL BIOLOGY CHAPTER 2 PRACTICE TEST QUESTIONS WITH CORRECT ANSWERS

Which of the following four statements, if any, is true regarding essential elements and living organisms? - Answer-Although all forms of life require iron, other elements are required only by certain species.

Consider this statement: Concentrations of trace elements in an environment above what is required by organisms generally have no effect on those organisms. Which of the following is a valid response to this statement? - Answer-False. Elevated concentrations of trace elements such as chromium and phosphorus can be toxic

Consider this statement: Oxygen is not a greenhouse gas; therefore, gases containing oxygen—such as ozone, nitrous oxide, and carbon dioxide—are not greenhouse gases either. Which of the following is a valid response to this statement? - Answer-False. Compounds can have emergent properties that are very different from those of the elements that form them

There are \_\_\_\_\_ naturally occurring elements. (eText Concept 2.1) - Answer-92

Which of the following is a trace element? (eText Concept 2.1) - Answer-copper

What are the four most abundant elements found in living systems? (eText Concept 2.1) - Answer-hydrogen, oxygen, nitrogen, and carbon

Which of the following has negligible mass? (eText Concept 2.2) - Answer-electron

Which of the following subatomic particles has appreciable mass and lacks a charge? (eText Concept 2.2) - Answer-neutron

The number of protons in an uncharged atom \_\_\_\_\_. (eText Concept 2.2) - Answer-equals the number of electrons

An element has eight protons, nine neutrons, and eight electrons. Its atomic number and atomic mass, respectively, are \_\_\_\_\_. (eText Concept 2.2) - Answer-8 and 17

An uncharged atom of nitrogen (atomic number = 7) has \_\_\_\_\_. (eText Concept 2.2) - Answer-seven protons and seven electrons

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