

Structure of H_2O
hydrogen bonding
Dissociation of H_2O



unit one guide

biological molecules
proteins
nucleic acids

H_2O properties: cohesion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, pH buffer, temperature stability

hydrogen bonds require lots of energy to change state of matter

COHESION → when H_2O bonds to it produces hydrogen and hydroxyl ions, H_2O dissociates, pH → more hydrogen
Adhesion → attracted to some surface
Surface Tension → hydrogen bonds
High specific heat → surface of the liquid doesn't heat

PHS bond: when bond → water expands due to hydrogen bonding, molecules move slowly

C	H	N	O	P	S
Carbon	Hydrogen	Nitrogen	Oxygen	Phosphorus	Sulfur

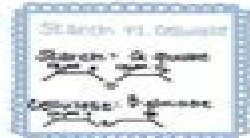
more hydrogen bonds → C → build biological molecules, cell formation
N → build proteins, nucleic acids
P → build nucleic acids + lipids
→ make up 98% of organisms

Carbohydrates	Lipids	Nucleic acids	Proteins
↓	↓	↓	↓
metabolites	energy	information	enzymes
↓	↓	↓	↓
C, H, O	C, H, O	C, H, N, O, P	C, H, N, O

Condensation synthesis → if the removal of an H_2O molecule to bond monomers
monomers to polymers, a bond between monomers (condensation) (polymerization)

Nucleic acid Structure:

* first carbon sugar (deoxyribose/ribose), phosphate, nitrogenous base (G, C, A, T), exact sequence of the code



Proteins: Specific order of amino acids (primary structure) → function

1° 1 group: side chains determine structure + function

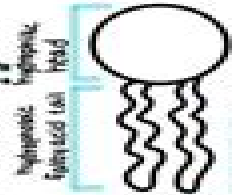


PROTEINS

* amino acids bond via a peptide bond (covalent) through dehydration synthesis
* 1 group change can cause a hydrophobic protein to lose the place of a hydrophilic protein

membranes

* made up of phospholipids
* create selectively permeable barrier
* contain phospholipids, cholesterol, and proteins



Amino acids:

* linear chains, connected by covalent bonds
* amino end and carboxyl end

NUCLEIC ACIDS

* add new nucleotides to the 3' end
* two nitrogenous bases: purine + pyrimidine
* purine is single ring, pyrimidine is double
* antiparallel double helix
* 5' on phosphate end, 3' on hydroxyl end
* both back and forth have nitrogenous bases (phosphodiester) → sugar-phosphate backbone

DEHYDRATION SYNTHESIS



covalent bonds form to create nucleic acids
* the number synthesizes the structure of code

ANTIPARALLEL



Chaperones:

* protein molecules that assist protein folding

PROTEIN STRUCTURE

Primary: Sequence of amino acids, specific order determines structure
Secondary: how folding of chain into alpha helix and beta pleated sheet
Tertiary: 3-D shape, hydrogen bonds, disulfide bond, ionic bond, hydrophobic int.
Quaternary: interactions between multiple polypeptide units

+ FUNCTION

* Enzymes → speeds up reaction rates in the cell
* Cell structure → part of the cellular membrane
* Communication between cells

DENATURATION:

* the protein starts to lose its shape/structure when (bonds break)
* temperature change
* pH of surrounding area
* salt concentration

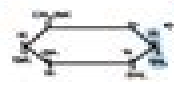
NH_2 = Amino
 $COOH$ = Carboxyl

ALDOSE VS

* contains aldehyde functional group

KETOSE

* contains ketone functional group
 $C=O$



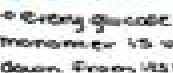
ALPHA VS

* all monomers are in the same organization



BETA

* Every glucose monomer is upside down from its neighbor



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web consequently the intosai auditing standards and indeed any auditing standards external to the sai cannot be prescriptive
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