

TESTS

EXPLORE CHEMISTRY WITH CHEMISTRY

DESCRIBING CHEMICAL REACTIONS

SOLUTIONS TO THE MODULE 7 REVIEW QUESTIONS

1. Elements cannot undergo decomposition reactions because elements have nothing to decompose into.
2. a. This equation represents a synthesis reaction because 2 elements come together to produce 1 compound.
b. This equation represents a formation reaction because 2 elements come together to produce 1 compound.
3. None of these
4. None of these
5. This equation represents a decomposition reaction because a single compound is breaking down into its elements.
6. The only equation that does not represent a formation reaction is (b). The equation for (b) is a formation equation because it starts with 2 reactants and makes only 1 in the end. Remember, reactions do not have to start with elements in order to be formation equations; we just assume they do when we write them ourselves.
7. The only equation that does not represent a combustion reaction is (b). The equation for (b) is a decomposition equation. Even though it produces CO_2 and H_2O , it does not have O_2 as a reactant, so it is not combustion.
8. Complete combustion produces carbon dioxide and water. Incomplete combustion produces carbon monoxide and water or carbon and water.

Apologia Chemistry Solutions And Test Manual

Jay L. Wile



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Apologia Chemistry Solutions And Test Manual Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Apologia Chemistry Solutions And Test Manual**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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