Atoms and molecules

an atom and its structure

Atoms are made up of particles of matter known as subatomic or fundamental particles. The three main subatomic particles are called protons, neutrons and electrons. An atom consists mostly of empty space. Almost all of the atom's mass is concentrated in a structure at its centre called the nucleus. The nucleus consists of protons and neutrons grouped together. Electrons orbit the nucleus.

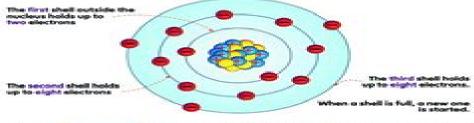
A proton is a positively charged particle A market has no electric charge Protons and payarons have almost the same masses

An electron is a negatively charged particle. It is kept in orbit around the nucleus because it is attracted to the positively charged protons. This attraction holds the atom together. An electron's mass is approximately 1840 times smaller than the mass of a proton or neutron.

An atom usually contains an equal number of positively charged protons and negatively charged electrons. This makes it electrically newtrol.

electrons and their energy levels

The way in which an atom reacts with other atoms (its chemical properties) is determined by the configuration of its electrons. Atoms can gain, lose or share electrons when they combine to form molecules. The electrons in an atom exist in layers called energy levels or shells. Electrons are arranged in the energy levels according to certain rules:





The attention number of an element tells you have many electrons that element has, An elem has as many electrons that it resets to orrange its electrons. The number of electrons is the souter energy level determines have received the orders and the souter energy level determines to be full. The most stable often order energy level determines to be full. The most stable adones are those with full subset many levels, like helium. The most reactive observa, like sodium, ore those that only need to love or goin over electron to obtain a full outer shell.

differences between mixtures & compounds

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"the components can be elements or compounds

The components are loosely put togeth

the properties of each part do not change.

the properties of the mixture the properties of all its ports.

COMPOUNDS (say, free supplies)

"pure substances." "the proportions of the ports are constant.

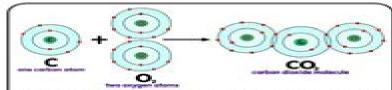
the core of

the properties of the compound are un-from the properties of the elements forming it, and from a mixture of these elements.

The mixture formed when one substance dissolves into another is called a **solution**. Sugar that is dissolved in water is a solution. There are always two parts of a solution: the solute and the solvent. The component which forms the greatest part of the solution is the solvent. Water is the solvent of the sugar solution. The sugar is the solution.

molecules

While an atom is the smallest particle of an element, molecules are the smallest particles of a compound. A molecule is formed when at least two atoms combine. These may be of the same element, or of different elements. If the atoms in a molecule are of different elements, it is called a compound. An example of such a compound is the carbon dicalds molecule:



A carbon atom has 4 electrons on its outer shell, while an oxygen atom has 6 electrons on its outer shell. It is thus necessary for 2 oxygen atoms to bond with 1 carbon atom so that the 2 oxygen atoms can fill their outer shells with 2 electrons, respectively, from the carbon atom ensuring that both oxygen atoms have full outer shells (8 required for 2nd shell) as does the carbon atom (2 required foor had absolit.

compounds & formulae

Compounds are formed when different elements combine. Like elements, each compound has a name and a symbol. The symbol of a compound is made of the symbols of the elements which form it, and is called a formula. A formula is used to show the atoms in a molecule. The numbers in a formula give the number of atoms of each kind of element in the molecule. The number of atoms of each element in that molecule is written after the symbol of the element, in small letters below the line e.g. O' which is a molecule of oxygen. A number written immediately before the formula of a molecule tells you how many molecules there are.







Atoms And Molecules A Childs Guide To Chemistry

Catharine Bomhold, Terri Elder

Atoms And Molecules A Childs Guide To Chemistry:

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three principles that all aspects of chemical education should be associated with research that the development of opportunities for chemical education should be both a continuous process and be linked to research and that the professional development of all those associated with chemical education should make extensive and diverse use of that research It is intended for pre service and practising chemistry teachers and lecturers chemistry teacher educators chemical education researchers the designers and managers of formal chemical curricula informal chemical educators authors of textbooks and curriculum support materials practising chemists and chemical technologists It addresses the relation between chemistry and chemical education curricula for chemical education teaching and learning about chemical compounds and chemical change the development of teachers the development of chemical education as a field of enquiry This is mainly done in respect of the full range of formal education contexts schools universities vocational colleges but also in respect of informal education contexts books science centres and museums **Research in Education** ,1969 **Subject Guide to** Children's Books in Print 1997 Bowker Editorial Staff, R Bowker Publishing, 1996-09 Build It, Make It, Do It, Play It! Catharine Bomhold, Terri Elder, 2014-06-30 A valuable one stop guide to collection development and finding ideal subject specific activities and projects for children and teens For busy librarians and educators finding instructions for projects activities sports and games that children and teens will find interesting is a constant challenge This guide is a time saving one stop resource for locating this type of information one that also serves as a valuable collection development tool that identifies the best among thousands of choices and can be used for program planning reference and readers advisory and curriculum support Build It Make It Do It Play It identifies hundreds of books that provide step by step instructions for creating arts and crafts building objects finding ways to help the disadvantaged or engaging in other activities ranging from gardening to playing games and sports Organized by broad subject areas arts and crafts recreation and sports including indoor activities and games and so forth the entries are further logically organized by specific subject ensuring quick and Concepts of Matter in Science Education Georgios Tsaparlis, Hannah Sevian, 2013-07-09 Bringing together a easy use wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter Concepts of Matter in Science Education informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning progressions for teaching a particle model of matter the mental models of both students and teachers of the particulate nature of matter educational technology chemical reactions and chemical phenomena chemical structure and bonding quantum chemistry and the history and philosophy of science relating to the particulate nature of matter The book will benefit a wide audience including classroom practitioners and student teachers at every educational level teacher educators and researchers in science education If gaining the precise meaning in particulate terms of what is solid what is liquid and

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Atoms And Molecules A Childs Guide To Chemistry Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Atoms And Molecules A Childs Guide To Chemistry," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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