



7. Which of the following is the atomic number of the atom represented by the diagram on the left?

- A. 22
- B. 25
- C. 47
- D. 5

8. What is the total electrical charge for the nucleus of an atom with 5 protons, 2 neutrons, and 10 electrons?

- A. +12
- B. -10
- C. +5
- D. -2

9. Which of the three following atoms has the lowest atomic mass?



1



2



3

- A. atom 1
- B. atom 2
- C. atom 3
- D. They all have equal atomic mass.

10. Which correctly describes the charges of the subatomic particles of an atom?

- A. electron/negative, proton/neutral, neutron/positive
- B. proton/positive, neutron/neutral, electron/negative
- C. neutron/negative, neutron/neutral, electron/positive
- D. neutron/negative, proton/positive, electron/neutral

Atomic Structure Guided Practice Problems Ans

**Kenneth Raymond Miller, Prentice Hall
(School Division)**



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