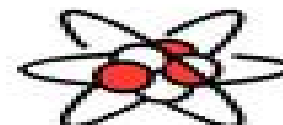


Name:

ATOMIC STRUCTURE

- 1 Use the words in the box to answer the following questions.

electron, neutron, nucleus, proton, shells



- (a) The central part of an atom is called the
- (b) The has a negative charge
- (c) The has no charge
- (d) The nucleus contains the and the
- (e) Electrons are found in
- (f) The and the have the same mass.

- 2 An atom contains 32 protons and 48 neutrons.

- (a) What is its atomic number?
- (b) How many electrons does it have?
- (c) What is the mass of the atom?

- 3 Complete the table to show the missing numbers of particles in the atoms A - E.

Atom	Protons	Neutrons	Electrons	Mass
A	20			40
B			21	45
C		24		44
D	19	21		
E			22	45

Which two atoms are isotopes of the same element? and

- 4 Complete the table to show the missing information:

Atom	Atomic number	Electron arrangement	Group in the Periodic Table	Period in the Periodic Table
P		2, 8, 7		
Q	11			
R			4	2
S			3	3

Atomic Structure Test Questions

M Lipman



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Table of Contents Atomic Structure Test Questions

1. Understanding the eBook Atomic Structure Test Questions
 - The Rise of Digital Reading Atomic Structure Test Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Test Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Test Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Test Questions
 - Personalized Recommendations
 - Atomic Structure Test Questions User Reviews and Ratings
 - Atomic Structure Test Questions and Bestseller Lists
5. Accessing Atomic Structure Test Questions Free and Paid eBooks

- Atomic Structure Test Questions Public Domain eBooks
- Atomic Structure Test Questions eBook Subscription Services
- Atomic Structure Test Questions Budget-Friendly Options
- 6. Navigating Atomic Structure Test Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Test Questions Compatibility with Devices
 - Atomic Structure Test Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Test Questions
 - Highlighting and Note-Taking Atomic Structure Test Questions
 - Interactive Elements Atomic Structure Test Questions
- 8. Staying Engaged with Atomic Structure Test Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Test Questions
- 9. Balancing eBooks and Physical Books Atomic Structure Test Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Test Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Test Questions
 - Setting Reading Goals Atomic Structure Test Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Test Questions
 - Fact-Checking eBook Content of Atomic Structure Test Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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