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web answers 1 hc 2h 3o 2 l 2o 2 g co 2 g 2h 2o l hc 2h 3o 2 l 4o 2 g 871 7 kj 2co 2 g 2h 2o l 2 the enthalpy of reaction is h
prod h react the reactant is at a higher potential than the product when the reaction occurs heat is released stabilizing the
system and the system moves to a lower energy potential 3

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web thermochemical equations quiz this online quiz is intended to give you extra practice in performing thermochemical
calculations with a variety of reactions including limiting reagents and percent yield options this quiz aligns with the
following ngss standard s hs ps1 4 hs ps3 1

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web in this set of practice questions we will summarize the main concepts of thermochemistry such as the relationship
between internal energy work and heat exothermic and endothermic process heat capacity constant pressure calorimetry
constant volume calorimetry the enthalpy the standard enthalpies of formation and their use in