

Strong vs. Weak Acids WebLab PhET

Go to: <https://phet.colorado.edu/en/simulation/acid-base-solutions> Click <Introduction> to begin.

Part 1: Procedure

1. The lab has 2 tools that allow you to test for pH values: A probe , and pH paper . Use each one by dipping it into the solution to be tested. Try all the given types of solutions and fill in the Data Chart with the pH value 0-14.

2. The circuit with a battery and bulb as shown:  is the tool used to test for conduction of a solution. By dipping the wire leads into the solution, the bulb will either remain unlit, be dimly lit, be somewhat bright or very bright. Test each solution and record your observation for the bulb's brightness in the chart below.

Part 1: Data	pH Value from Probe	Color & pH Value from pH Paper	Observations from Circuit Tool Describe the brightness
Water (H ₂ O) 	7.00	Yellow / orange 7	light
Strong Acid (HA) 	2.00	Red / orange 6	Very bright
Weak Acid (A) 	4.50	orange 6	Bright
Strong Base (MOH) 	12.00	Blue / green 11	Very bright
Weak Base (B) 	9.50	Blue / green 10	Bright

Part 1: Analysis

1. What pH value range is observed: a. for acids? 0 - 6 b. for bases? 8 - 14

2. Why are some solutions better conductors of electricity?

The strong electrolytes often supply ions to the solution. This happens more than the weak electrolytes. It is determined by the high and low pH levels.

Answer Key Pogil Strong Versus Weak Acids

Michal Rosen-Zvi



Answer Key Pogil Strong Versus Weak Acids:

Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more **ACID-BASE EQUILIBRIA** NARAYAN CHANGDER, 2024-05-16 Note Anyone can request the

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A Test of the Theory of the Effect of Strong Acids on the Solubility of the Salts of Weak Acids David Schwartz, 1897

Weak Acid - Strong Base Titration, 200? **Studying the Ph of Strong Acid, Weak Acid, Salt & Buffer Solutions**

Marcia L. Gillette, Norma Robinson North, 1998-01-01

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Weak Acids and Bases, 2016 Look at weak acids and bases compounds that are only slightly ionized in water based solutions. You learn how to solve the classic weak acid problem and apply the same approach to weak base problems.

Proton Chemistry Robert A. Richardson, 1976 Alberta Authorized Resource for grade 12 ca 1980 1994

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