

Name: \_\_\_\_\_

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55 multiple-choice questions (3 pts each) + 6 fill-in questions (40 pts) = Total (205 pts).

1) How many **unpaired** electrons are present in the electron configuration of the **O** atom?  
 A) none      B) one      **C) two**      D) three      E) four

2) Which of the following contain(s) **polar covalent** bonds?  
 A) H<sub>2</sub>      B) NaI      C) Cl<sub>2</sub>      **D) H<sub>2</sub>O**      E) both A and C

3) What is the formal charge and hybridization for the highlighted **Carbon**?  
 A) 0, sp<sup>2</sup>      B) +1, sp<sup>2</sup>      **C) -1, sp**      D) +1, sp      E) 0, sp



4) What of the following best describes the hybridization of C in **carbocation** CH<sub>3</sub><sup>+</sup>?  
 A) sp<sup>3</sup>      **B) sp<sup>2</sup>**      C) sp      D) s

5) what orbital is the nitrogen lone pair residing in the following molecule?  
 A) sp<sup>3</sup>      **B) sp<sup>2</sup>**      C) sp      D) p      E) s



6) Identify the **most acidic** carboxylic acid.



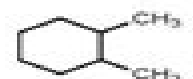
**Answer is E**

7) The **conjugate acid** of H<sub>2</sub>O is \_\_\_\_\_

A) HO<sup>-</sup>      B) H<sub>3</sub>O      **C) H<sub>3</sub>O<sup>+</sup>**      D) H<sub>2</sub>O      E) H<sub>2</sub>O<sup>+</sup>

8) How many potential stereoisomers exist for the following molecule.

A) 1      B) 2      **C) 3**      D) 4      E) 5



9) what is the name of the following compound?

A) *trans*-4-ethylcyclohexan-1-ol      B) *cis*-4-ethylhexan-1-ol  
 C) *trans*-4-methylcyclohexan-1-ol      **D) *cis*-4-ethylcyclohexan-1-ol**  
 E) *trans*-4-ethylcyclopentan-1-ol



10) which of the following step is part of the propagation in the methane chlorination?

A)  $\text{Cl}-\text{Cl} \longrightarrow 2 \text{Cl} \cdot$       B)  $\text{H}_3\text{C} \cdot + \text{Cl}-\text{Cl} \longrightarrow \text{H}_3\text{C}-\text{Cl} + \text{Cl} \cdot$   
 C)  $\text{H}_3\text{C}-\text{H} + \cdot\text{Cl} \longrightarrow \text{H}_3\text{C} \cdot + \text{H}-\text{Cl}$       D) B and C

**Answer is D**

11) Based on the Hammond postulate, which of the following transition state best explains the following reaction step?

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