

Student Exploration: Bohr Model: Introduction

When light passes through a gas, certain wavelengths of the light are absorbed. The result is a unique absorption spectrum. Two examples are shown below.



- What colors of light are absorbed by hydrogen gas? violet, blue, green, and red
- What colors of light are absorbed by helium gas? violet, blue, green, yellow, and red

Gizmo Warm-up

In 1913, Niels Bohr proposed that the unique spectral lines created by different elements were related to the way electrons were arranged around the nucleus. The Bohr Model: Introduction Gizmo explores this connection.

The laser shown in the Gizmo can emit photons, or particles of light, at a variety of wavelengths. The energy of a photon, measured in electron volts (eV), is inversely proportional to its wavelength. Photons that pass through the gas are detected by the photon detector at the right.



 With the Energy (eV) set to 1 eV, click Fire. Did the photon go straight through the gas in the tube, or was it absorbed by the gas?
the photon went straight through the gas tube.

Set the Energy (eV) to 4 eV, and click Fire. What happened this time? the photon was absorbed by the gas.

Activity A: Absorption spectra

Get the Gizmo ready:

- On the SIMULATION pane, select Lamp.
- Check that Gas C is selected.



Introduction: The smaller the wavelength of a photon, the greater its energy. Humans can see photons with wavelengths between 700 nanometers (red) and 400 nanometers (violet), corresponding to energies between 1.8 and 3.1 eV.

Question: What does the absorption spectrum of an element indicate about its electron configuration?



Bohr Gizmo Answers

Hendrik Anthony Kramers



Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Bohr Gizmo Answers**

In a world inundated with screens and the cacophony of instant communication, the profound energy and psychological resonance of verbal artistry often disappear in to obscurity, eclipsed by the continuous onslaught of noise and distractions. However, nestled within the musical pages of **Bohr Gizmo Answers**, a interesting function of literary splendor that impulses with fresh thoughts, lies an remarkable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, this enchanting opus instructions readers on a mental odyssey, softly revealing the latent potential and profound affect stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is central styles, dissect its captivating publishing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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