# PHYSICAL SCIENCE PRACTICE TEST WITH ANSWER KEY

 A week attraction between hydrogen atom in one molecule and an oxygen in other called\_\_\_\_\_.

A. covalent bond

B. Hydrogen bond

C. Ionic bond

D. Metallic bond

2. Which of the following substances is most likely a covalent compound ?

A. CaCO3

B. KNO3

C. SO2

D. NaOH

3. What is the element needed to bond with Na to create the most ionic bond?

A. F.

B. C1

C. Br

D. I

4. The boiling point of substance A is higher than that of substance B. What does this meaning?

A. Substance A is less viscous than substance B.

B. Substance A evaporates faster than substance B.

C. Substance B has greater surface tension than substance A.

D. Substance B has weaker intermolecular forces than substance A.

5. What is the intermolecular force of attraction between the nucleus of one atom and the electrons of a nearby atom?

A. Dipole-dipole force

B. Ion-dipole force

C. London dispersion force

D. Van der Waals force

6. Which of the following has the greatest number of atom?

A. Ca(OH)2

B. A12O3

## **Apexvs Answer Key Physical Science**

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