

ABAQUS

Version 6.6 Documentation

ABAQUS Analysis User's Manual

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18.2.2 Models for metals subjected to cyclic loading

Products: ABAQUS/Standard ABAQUS/Explicit ABAQUS/CAE

References

- ["Material library: overview," Section 16.1.1](#)
- ["Inelastic behavior," Section 18.1.1](#)
- ["Anisotropic yield/creep," Section 18.2.6](#)
- ["UHARD," Section 1.1.26 of the ABAQUS User Subroutines Reference Manual](#)
- [*CYCLIC HARDENING](#)
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Overview

The kinematic hardening models:

- are used to simulate the inelastic behavior of materials that are subjected to cyclic loading;
- include a linear kinematic hardening model and a nonlinear isotropic/kinematic hardening model;
- can be used in any procedure that uses elements with displacement degrees of freedom;
- in ABAQUS/Standard cannot be used in adiabatic analyses, and the nonlinear isotropic/kinematic hardening model cannot be used in coupled temperature-displacement analyses;
- can be used to model rate-dependent yield;
- can be used with creep and swelling in ABAQUS/Standard; and
- require the use of the linear elasticity material model to define the elastic part of the response.

Yield surfaces

The kinematic hardening models used to model the behavior of metals subjected to cyclic loading are pressure-independent plasticity models; in other words, yielding of the metals is independent of the equivalent pressure stress. These models are suited for most metals subjected to cyclic loading conditions, except voided metals. The linear kinematic hardening model can be used with the Mises or Hill yield surface. The nonlinear isotropic/kinematic model can be used only with the Mises yield surface in

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**Jie Yang, Jiyang Fu, Airong Liu, Ching-Tai
Ng**



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web mar 29 2016 chapter test a chapter chemical equations and reactions in the space provided write the letter of the term or phrase that best completes each statement or best answers each question 1 you mix solution a with solution b in a beaker which of the following observations does not help you prove that a chemical reaction has occurred

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web terms in this set 35 a chemical equation that does not indicate relative amounts of reactants and products skeleton equation a new substance formed in a chemical reaction product a starting substance reactant a concise representation of a chemical balanced equation

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web 1 20 flashcards test match q chat created by reesek24030 terms in this set 20 the production of one or more new substances the only sure evidence for a chemical reaction is inhibitor a material used to decrease the rate of a chemical reaction is a n matter is not created or destroyed

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web aug 29 2023 chemical reactions are part of our daily lives from cooking in the kitchen to driving a car these reactions are commonplace in a chemical reaction the molecules of one substance break apart and join together with those of another substance to create a different compound combination of molecules many chemical reactions are non