



ABSOLUTE RADIOMETRY

Edited by

F. Hengstberger

Absolute Radiometry Electrically Calibrated Thermal Detectors Of Optical Radiation

BM King



Absolute Radiometry Electrically Calibrated Thermal Detectors Of Optical Radiation:

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 pK_b of the weak acid or base being titrated a solution ph as a function of the volume of 1.00 M NaOH added to 10.00 ml of
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the word titration comes from the latin word titulus which means inscription or title

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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should
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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of HCl because this is a strong acid the
ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then $ph = -\log 0.100 = 1.000$ ph $-\log 0.100 = 1.000$ b titrant volume 12.50 ml

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reacting it with a standard solution one type of titration uses a neutralization reaction in which an acid and a base react to
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web expert answer 100% 2 ratings the example is based upon Henderson-Hasselbalch equation the equation gives relation

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Volume of NaOH added (mL)	pH of the solution
4.6	4.6
4.6	5.9
5.9	5.9

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