
Test Bank for Fundamentals of Biochemistry Life at the Molecular Level 5th Edition by Donald Voet, Judith G. Voet, Charlotte W. Pratt

1. Which of the following is a characteristic of water that makes it essential for life?

- a) Water has a high vapor pressure.
- b) Water is a non-polar solvent.
- c) Water has a high heat capacity.
- d) Water has a low dielectric constant.
- **Answer: c) Water has a high heat capacity.**

2. Which of the following types of bonds is most important for the stabilization of protein secondary structures such as alpha helices and beta sheets?

- a) Disulfide bonds
- b) Hydrogen bonds
- c) Ionic bonds
- d) Covalent bonds
- **Answer: b) Hydrogen bonds.**

3. In the Michaelis-Menten equation, which assumption is made regarding the enzyme-substrate complex (ES)?

- a) The concentration of ES is constant during the reaction.
- b) ES is rapidly converted to enzyme and product.
- c) The enzyme is saturated with substrate at all concentrations.
- d) The enzyme binds only a single substrate.
- **Answer: a) The concentration of ES is constant during the reaction.**

4. Which of the following amino acids has a side chain that can form hydrogen bonds and is often found in enzyme active sites?

- a) Alanine
- b) Glycine
- c) Serine
- d) Valine
- **Answer: c) Serine.**

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