

Structure of H_2O
hydrogen bonding
Dissociation of H_2O



unit one guide

biological molecules
proteins
nucleic acids

H_2O properties: cohesion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, pH buffer, temperature stability

hydrogen bonds require lots of energy to change state of matter

COHESION → when H_2O bonds to it produces hydrogen and hydroxyl ions, H_2O dissociates, pH → more hydrogen
Adhesion → attracted to some surface
Surface Tension → hydrogen bonds
High specific heat → surface of the liquid doesn't heat

PHS bond when bond → water expands due to hydrogen bonding, molecules move slowly

C
Carbon

H
Hydrogen

N
Nitrogen

O
Oxygen

P
Phosphorus

S
Sulfur

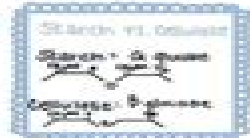
more hydrogen bonds
C → build biological molecules, cell formation
N → build proteins, nucleic acids
P → build nucleic acids + lipids
→ make up 98% of organisms

Carbohydrates ↓ monosaccharides C, H, O	Lipids ↓ fats + oils C, H, O	Nucleic acids ↓ monosaccharides C, H, N, O, P	Proteins ↓ amino acids C, H, N, O
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Condensation synthesis → if the removal of an H_2O molecule to bond monomers. Hydrolysis is the addition of a water molecule to cleave a bond between monomers (breaks protein)

Nucleic acid Structure:

* first carbon sugar (deoxyribose/ribose), phosphate, nitrogenous base (G, C, A, T), exact shape of strand



Proteins: Specific order of amino acids. Primary structure → 1° group. Substructure determines structure + function

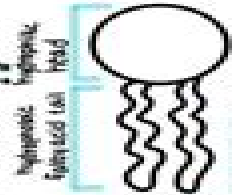


PROTEINS

* amino acids bond via a peptide bond (covalent) through dehydration synthesis
* 1° group changes can cause a hydrophobic protein to lose the place of a hydrophilic protein

membranes

* made up of phospholipids
* create selectively permeable barrier
* contain phospholipids, cholesterol, and proteins



Amino acids:

* linear chains, connected by covalent bonds
* amino end and carboxyl end

NUCLEIC ACIDS

* add new nucleotides to the 3' end
* two nitrogenous bases: purine + pyrimidine
* purine is single ring, pyrimidine is double
* antiparallel double helix
* 5' on phosphate end, 3' on hydroxyl end
* both back and forth have nitrogenous bases perpendicular to sugar-phosphate backbone

DEHYDRATION SYNTHESIS



covalent bonds form to create nucleic acids
* the number synthesizes the structure of order

ANTIPARALLEL



Chaperonins:

* protein molecules that assist protein folding

PROTEIN STRUCTURE

Primary: Sequence of amino acids, specific order determines structure
Secondary: how folding of chain into alpha helix and beta pleated sheets
Tertiary: 3-D shape, hydrogen bonds, disulfide bond, ionic bond, hydrophobic int.
Quaternary: interactions between multiple polypeptide units

+ FUNCTION

* Enzymes → speeds up reaction rates in the cell
* Cell structure → part of the cellular membrane
* Communication between cells

DENATURATION:

* the protein starts to lose its shape/structure when bonds break
* temperature changes
* pH of surrounding area
* salt concentration

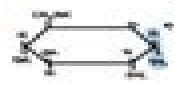
NH_2 = Amino
 $COOH$ = Carboxyl

ALDOSE VS

* contains aldehyde functional group

KETOSE

* contains ketone functional group
 $C=O$



ALPHA VS BETA

* all monomers are in the same organization



* Every glucose monomer is upside down from its neighbor

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JA Banks



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1. Understanding the eBook Biology Study Guide All
 - The Rise of Digital Reading Biology Study Guide All
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology Study Guide All
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology Study Guide All
 - User-Friendly Interface
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 - Biology Study Guide All and Bestseller Lists
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 - Highlighting and Note-Taking Biology Study Guide All
 - Interactive Elements Biology Study Guide All
- 8. Staying Engaged with Biology Study Guide All
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology Study Guide All
- 9. Balancing eBooks and Physical Books Biology Study Guide All
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology Study Guide All
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Biology Study Guide All
 - Setting Reading Goals Biology Study Guide All
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Biology Study Guide All
 - Fact-Checking eBook Content of Biology Study Guide All
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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