

1. When an electron is removed from a molecule, the molecule is _____ **oxidized** _____ and the amount of potential energy in the molecule (**increases/decreases**). When an electron is added to a molecule, it is said to be _____ **reduced** _____ and the amount of potential energy in the molecule (**increases/decreases**).
2. When you track energy transfers through a biological system, you are tracking the path of _____ **electrons** _____.
3. The principal electron carriers in cellular respiration are derivatives of _____ **nucleotides** _____.
4. NAD is derived from _____ **vitamin B₃, niacin** _____. The oxidized form is _____ **NAD⁺** _____ and the reduced form is _____ **NADH** _____.
5. FAD is derived from _____ **vitamin B₂, also called riboflavin** _____.
6. Why can a cell not store free energy for use, but must instead rely on ATP?

Excess free energy would result in an increase of heat in the cell, which would result in excessive thermal motion that could damage and then destroy the cell. Rather, a cell must be able to handle that energy in a way that enables the cell to store energy safely and release it for use only as needed.

7. In nearly every living organism, the energy needed to generate ATP predominantly comes from _____ **glucose** _____, _____ **fructose** _____, and _____ **galactose** _____.
8. _____ **Phosphorylation** _____ is the addition of a phosphate group to another molecule.

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Gerald L. Kellogg, Starr



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Table of Contents Biology Chapter 51 Guided Assignment Answers

1. Understanding the eBook Biology Chapter 51 Guided Assignment Answers
 - The Rise of Digital Reading Biology Chapter 51 Guided Assignment Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology Chapter 51 Guided Assignment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology Chapter 51 Guided Assignment Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology Chapter 51 Guided Assignment Answers
 - Personalized Recommendations
 - Biology Chapter 51 Guided Assignment Answers User Reviews and Ratings
 - Biology Chapter 51 Guided Assignment Answers and Bestseller Lists
5. Accessing Biology Chapter 51 Guided Assignment Answers Free and Paid eBooks
 - Biology Chapter 51 Guided Assignment Answers Public Domain eBooks
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 - Biology Chapter 51 Guided Assignment Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Biology Chapter 51 Guided Assignment Answers Compatibility with Devices
 - Biology Chapter 51 Guided Assignment Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology Chapter 51 Guided Assignment Answers
 - Highlighting and Note-Taking Biology Chapter 51 Guided Assignment Answers
 - Interactive Elements Biology Chapter 51 Guided Assignment Answers
8. Staying Engaged with Biology Chapter 51 Guided Assignment Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology Chapter 51 Guided Assignment Answers
9. Balancing eBooks and Physical Books Biology Chapter 51 Guided Assignment Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology Chapter 51 Guided Assignment Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Biology Chapter 51 Guided Assignment Answers
- Setting Reading Goals Biology Chapter 51 Guided Assignment Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biology Chapter 51 Guided Assignment Answers
- Fact-Checking eBook Content of Biology Chapter 51 Guided Assignment Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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