

PHYSICAL PHARMACEUTICS

QUESTIONS:

1. Solubility of gases increases with decrease of:
a) Mass. b) Volume. c) Temperature. d) Pressure.
2. When a large amount of solute dissolves in a given amount of solvent it forms:
a) Dilute solution.
b) Saturated solution.
c) Aqueous solution.
d) Concentrated Solution.
3. Solution which can hold no more solute is called:
a) Dilute solution.
b) Saturated solution.
c) Aqueous solution.
d) Concentrated Solution.
4. Which is not a colligative property:
a) Raoult's law.
b) Depression in freezing point.
c) Osmotic pressure.
d) Melting point.
5. Azeotropic mixtures are:
a) Constant temperature boiling mixtures.
b) Those which boils at different temperatures.
c) Mixture of two solids.
d) None of above.
6. The pH of pharmaceutical buffer system can be calculated by:
a) pH partition theory.
b) Noyes Whitney law.
c) Henderson - Hasselbalch equation.
d) Michaelis Menten equation.
7. Unit of surface tension is:
a) Dyne/meter
b) Dyne/cm
c) Cg/cm
d) None of the above.
8. Two solutions are said to be isotonic if they exert same.....
a) Viscosity. c) Osmotic pressure.
b) Surface tension. d) None of the above.
9. Dissolution is affected by:
a) Surface area.
b) Viscosity.
c) Temperature.
d) All of the above.

Basic Physical Pharmacy Questions With Answer

Manan Shroff



Basic Physical Pharmacy Questions With Answer:

Basic Physical Pharmacy Joseph K. H. Ma, Boka Hadzija, 2013 Basic Physical Pharmacy provides a thorough yet accessible overview of the principles of physical pharmacy and their application in drug formulation and administration. This definitive guide to physical pharmacy covers all types of pharmaceuticals from traditional forms and dosages to nanotechnology based novel dosage design. Authored by two nationally recognized pharmaceutical scientists and active pharmacy faculty, Basic Physical Pharmacy is clearly organized into four sections: Physical Pharmacy in Solutions, Solid Dosage Forms, Polyphasic Systems, and Drug Delivery and Novel Drug Delivery Systems. Students can build upon their chemistry education to learn the physicochemical properties of drugs and their therapeutic effects on the body. With a highly accessible approach, Basic Physical Pharmacy will help students comprehend and apply the principles of physical pharmacy in clinical practice. Covers major drug products and delivery systems. Features current trends in pharmaceutical research and development including nanotechnology based dosage design. Includes many examples of useful equations and formulation methods. Contains over 200 illustrations, photos, and tables. Topics include: Solutions, Ionization of Drugs in Solutions, Buffers and Buffered Solutions, Drug Solubility, Diffusion and Dissolution, Distribution Phenomena, Complexation and Protein Binding, Interfacial Phenomena, Rheology, Colloids, Suspensions and Emulsions, Semisolid Dosage Forms, Dermatologicals, Suppositories, Powders, Capsules, Tablets, Aerosols, Sterile Dosage Forms, Ophthalmic Formulations, Radiopharmaceuticals, Modified Release Drug Delivery Systems, Biotechnology Products, Drug Product Stability. Each new print textbook includes an access code for the online Companion Website. Ebooks do not include access to the Companion Website. Access to the Companion Website may also be purchased separately under the RESOURCES tab. FOR STUDENTS: Student Companion Website includes Cross Words, Flash Cards, Interactive Glossary, Matching Questions, Instructor Resources, Answers to End of Chapter Questions, Image Bank, Power Point Presentations, Test Bank.

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Table of Contents Basic Physical Pharmacy Questions With Answer

1. Understanding the eBook Basic Physical Pharmacy Questions With Answer
 - The Rise of Digital Reading Basic Physical Pharmacy Questions With Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Basic Physical Pharmacy Questions With Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Basic Physical Pharmacy Questions With Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Basic Physical Pharmacy Questions With Answer
 - Personalized Recommendations
 - Basic Physical Pharmacy Questions With Answer User Reviews and Ratings
 - Basic Physical Pharmacy Questions With Answer and Bestseller Lists
5. Accessing Basic Physical Pharmacy Questions With Answer Free and Paid eBooks
 - Basic Physical Pharmacy Questions With Answer Public Domain eBooks
 - Basic Physical Pharmacy Questions With Answer eBook Subscription Services
 - Basic Physical Pharmacy Questions With Answer Budget-Friendly Options
6. Navigating Basic Physical Pharmacy Questions With Answer eBook Formats

- ePub, PDF, MOBI, and More
- Basic Physical Pharmacy Questions With Answer Compatibility with Devices
- Basic Physical Pharmacy Questions With Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Basic Physical Pharmacy Questions With Answer
 - Highlighting and Note-Taking Basic Physical Pharmacy Questions With Answer
 - Interactive Elements Basic Physical Pharmacy Questions With Answer
- 8. Staying Engaged with Basic Physical Pharmacy Questions With Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Basic Physical Pharmacy Questions With Answer
- 9. Balancing eBooks and Physical Books Basic Physical Pharmacy Questions With Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Basic Physical Pharmacy Questions With Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Basic Physical Pharmacy Questions With Answer
 - Setting Reading Goals Basic Physical Pharmacy Questions With Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Basic Physical Pharmacy Questions With Answer
 - Fact-Checking eBook Content of Basic Physical Pharmacy Questions With Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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