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## The Hydrogen Atom - Student Guide

#### **Background Material**

Carefully read the background pages entitled Energy Levels, Light, and Transitions and answer the following questions to check your understanding.

Question 1: Complete the following table which compares how the Bohr Model and the Quantum model represent the Hydrogen atom. In some cases they both describe things in the same way and in some cases they do not.

Concept	Bohr Model	Quantum Model	
Nucleus	positively charged proton at the center	positively charged proton at the center	
Electrons	Negatively charged cloud	negatively charged cloud	
Orbital	Electrons travel in a circular orbits around nucleus	Electrons are in a certain position in orbit	
Location of electron in the first orbital	always located at the Bohr radius	Always located on electron cloud	
Energy of the Energy Levels	80-8642	$E = -E_0/rr^2$	

Question 2: Complete the following table which relates the parameters of two different photons by circling the appropriate relationship. The first row is completed for you: "A red photon has a larger wavelength, smaller frequency, smaller energy, and the same velocity through space as a blue photon".

# **Answers To The Hydrogen Atom Student Guide**

David R. Klein

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authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters. The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 the text Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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