Name: _

Date:

STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:

Calculate how many grams of iodine are needed to prepare 28.6 grams of ICI by this reaction.

2. Using the following equation:

$$5 \text{ KNO}_2 + 2 \text{ KMnO}_4 + 3 \text{ H}_2 \text{SO}_4 \rightarrow 5 \text{ KNO}_3 + 2 \text{ MnSO}_4 + \text{ K}_2 \text{SO}_4 + 3 \text{ H}_2 \text{O}_4$$

How many moles and grams of $KMnO_4$ are needed for this reaction on 11.4 grams of KNO_2 ?

3. Using the following equation:

How many moles and grams of oxygen (O_2) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:

$$NaIO_3 + 6 HI \rightarrow 3 I_2 + NaI + 3 H_2O$$

Calculate the number of moles and the number of grams of iodine (l_2) that can be made this way from 16.4 grams of NaIO₃.

Basic Stoichiometry Practice Problems

David A. Ucko

Basic Stoichiometry Practice Problems:

Enjoying the Song of Expression: An Psychological Symphony within **Basic Stoichiometry Practice Problems**

In a global consumed by monitors and the ceaseless chatter of quick connection, the melodic elegance and mental symphony created by the written term often diminish into the back ground, eclipsed by the persistent sound and disruptions that permeate our lives. However, nestled within the pages of **Basic Stoichiometry Practice Problems** an enchanting literary prize filled with raw emotions, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, this interesting masterpiece conducts viewers on an emotional trip, well unraveling the hidden songs and profound influence resonating within each carefully constructed phrase. Within the depths of the touching examination, we shall explore the book is central harmonies, analyze its enthralling publishing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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