

Introduction

This exam is a closed-book, closed-notes examination. You are permitted to bring a calculator and a ruler to the exam. All other materials, including mobile phones, are prohibited. The exam is 1 hour long. The questions are multiple choice. You are to select the best answer for each question. The questions are based on the material covered in the course. The questions are based on the material covered in the course. The questions are based on the material covered in the course.

Instructions: Read the questions carefully. Choose the best answer for each question. Write your answer in the space provided. Do not write in the spaces between the questions. Do not write on the back of the page. Do not use a calculator or ruler.

Part 1: Multiple Choice Questions



1. Which of the following is a physical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Boiling point
2. Which of the following is a chemical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Flammability



3. Which of the following is a physical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Boiling point
4. Which of the following is a chemical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Flammability

Question	Answer A	Answer B	Answer C	Answer D	Answer E
1.	Color	Mass	Density	Boiling point	Flammability
2.	Color	Mass	Density	Boiling point	Flammability
3.	Color	Mass	Density	Boiling point	Flammability
4.	Color	Mass	Density	Boiling point	Flammability

5. Which of the following is a physical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Boiling point
6. Which of the following is a chemical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Flammability

Part 2: Multiple Choice Questions



7. Which of the following is a physical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Boiling point
8. Which of the following is a chemical property of a substance?
 - a. Color
 - b. Mass
 - c. Density
 - d. Flammability



10. Complete the table below with the missing values for the given substances. Use the periodic table.

Element (Symbol)	Atomic Number (Z)	Atomic Mass (A)	Group	Period
H (Hydrogen)	1	1.008	1	1
He (Helium)	2	4.003	18	1
Li (Lithium)	3	6.941	1	2
Be (Beryllium)	4	9.012	2	2

Write the answers in the spaces provided. Do not write on the back of the page.

Basic Stoichiometry Phet Lab Answers

Christian G. Meyer



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