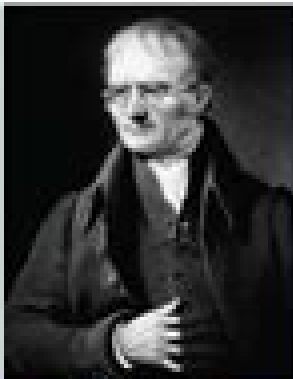
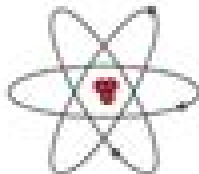
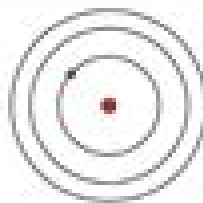
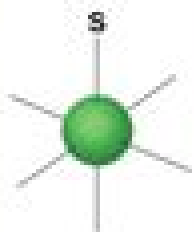


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Atomic And Quantum Theory In Chemistry

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Atomic and Quantum Physics Hermann Haken, Hans C. Wolf, 2012-12-06 Atomic physics and its underlying quantum theory are the point of departure for many modern areas of physics astrophysics chemistry biology and even electrical engineering This textbook provides a careful and eminently readable introduction to the results and methods of empirical atomic physics The student will acquire the tools of quantum physics and at the same time learn about the interplay between experiment and theory A chapter on the quantum theory of the chemical bond provides the reader with an introduction to molecular physics Plenty of problems are given to elucidate the material The authors also discuss laser physics and nonlinear spectroscopy incorporating latest experimental results and showing their relevance to basic research Extra items in the second edition include solutions to the exercises derivations of the relativistic Klein Gordon and Dirac equations a detailed theoretical derivation of the Lamb shift a discussion of new developments in the spectroscopy of inner shells and new applications of NMR spectroscopy for instance tomography

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Relativistic Quantum Theory of Atoms and Molecules Ian P Grant, 2007-04-15 This book is intended for physicists and chemists who need to understand the theory of atomic and molecular structure and processes and who wish to apply the theory to practical problems As far as practicable the book provides a self contained account of the theory of relativistic atomic and molecular structure based on the accepted formalism of bound state Quantum Electrodynamics The author was elected a Fellow of the Royal Society of London in 1992

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structure and spectroscopy This book provides a gentle introduction to the subject with the required background in physics and mathematics kept to a minimum It develops the basic concepts needed as background The emphasis throughout is on the physical concepts and their application in chemistry especially to atoms and to the periodic table of elements The Quantum Theory of Atoms in Molecules Chérif F. Matta, Russell J. Boyd, 2007-04-09 This book distills the knowledge gained from research into atoms in molecules over the last 10 years into a unique handy reference Throughout the authors address a wide audience such that this volume may equally be used as a textbook without compromising its research oriented character Clearly structured the text begins with advances in theory before moving on to theoretical studies of chemical bonding and reactivity There follow separate sections on solid state and surfaces as well as experimental electron densities before finishing with applications in biological sciences and drug design The result is a must have for physicochemists chemists physicists spectroscopists and materials scientists *Atomic and Quantum Physics* H Haken, H C Wolf, W D Brewer, 1984-07-01 *The Physics of Atoms and Quanta* Hermann Haken, Hans C. Wolf, 2012-12-06 This fourth edition contains a few additional figures Otherwise only typographical errors have been removed The final chapter on Fundamentals of the Quantum Theory of Chemical Bonding is continued in an extended way in the textbook *Molecular Physics and Elements of Quantum Chemistry* by the same authors This book contains in particular a profound presentation of group theory as applied to atoms and molecules Furthermore the interaction between atoms and molecules and light is treated in detail We thank again Springer Verlag in particular Dr H 1 Kblsch and Mr C D Bachem for their excellent cooperation as always and Prof W D Brewer for his continuous support in translating our German text Stuttgart February 1994 H Haken H C Wolf Preface to the Third Edition The second edition of this book again enjoyed a very positive reception from both university teachers and students In this edition we have removed all of the typographical errors that came to our attention In order to keep the book as current as possible new developments in the direct observation of individual atoms in electromagnetic traps Paul traps and of atoms in molecules on solid surfaces using the scanning tunnel microscope have been added to this edition **The Physics of Atoms and Quanta** Hermann Haken, Hans Christoph Wolf, 2006-05-24 The highly positive affirmation and wide reception that this book continues to receive from professors and students alike is the occasion for this 7th edition Once again we have included a number of valuable suggestions for improvements which we address as appropriate In addition we refer to a number of developments in atomic physics Of these new developments in regard to exotic atoms we mention antihydrogen in particular because fundamental experiments in matter and antimatter can be expected in the future Furthermore we have inserted a chapter on the behaviour of atoms in strong electrical fields Experiments with corresponding lasers could only recently be realized We thank our Jenaer colleague R Sauerbrey for his contribution of this chapter We have also included a new chapter on the behaviour of the hydrogen atom in strong magnetic fields The results are of profound interest for two very different fields of physics on the one hand according to classical

physics one expects chaotic behaviour from Rydberg atoms in magnetic fields that can be created in the laboratory thus an association can be drawn to aspects of chaos theory and the problems of quantum chaos On the other hand the very strong fields necessary for low quantum numbers are realized in the cosmos in particular with white dwarfs and neutron stars

Quantum Chemistry of Atoms and Molecules Philip S. C. Matthews, 1986-12-11 In this 1986 book Dr Matthews emphasises the fundamental ideas of quantum theory as they relate to mainstream areas of quantum theory such as bonding and spectroscopy elementary ideas on the use of symmetry are also included No prior knowledge of quantum theory is assumed and considerable help is given in understanding the mathematics that is involved Much of this mathematics is integrated into the text but the more intricate portions are to be found in separate boxes these can be left on a first reading and returned to later There are also questions and problems at the end of nearly every section and these are designed to test the student's understanding of the text and to give fresh insights into the work full answers are provided

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