Word Bank:

Acidic

Acids Atoms

Blood

Blue

Buffers Carbon dioxide

> Center Dissolve

Electrons

Ends Environment

Gained Hydrogen

Hydroxyl Indicator

Neutral Neutralize

pH

Red

Salts Scale Shares Stomach acid

> Strong Water

Weak

Visit the main page of the Kid Zone at http://sciencespot.net/ to find the link for BrainPOP.

- The term ______ refers to the chemical potential of hydrogen.
 The pH _____ measures how _____ or basic a substance.
- The pH ____ measures how ____ or basic a substance on a scale of 1 to 14.

 ____ measure from 1 to 7, while the ____ or alkaline side measures from 7 to 14.
- Identify each as an acid (A) or a base (B).

____Lemon juice ____Soap ____Battery acid ____Hydrochloric acid ____Toothpaste ____Yinegar ___Baking soda ____Floor cleaner

Really strong acids and bases are found at the ______ of the scale, while those near the ______ are weaker. Right in the middle at exactly 7, or pH ______, you have pure water.

5. Acids contain _______ ions (H+), while bases contain _______.

(OH-) ions. Ions are atoms that have either lost or gained _______. In bases you find hydroxyl ions, which have _______ electrons. Hydrogen ions found in acids are ______ that have had an electron knocked off, which is why these ions are always trying to get the lost electrons back. When you put a metal in acid, the metal starts to

6. Bases have the power to ______ acids. The hydroxyl ion _____ its electron with the hydrogen ion. The H+ from the hydrogen ion and the OH- from the hydroxyl ion bond together to form H2O, which is _____ with pH neutral.

Different types of _____ are formed depending on the types of acids and bases involved in the reaction.

When Moby poured the baking soda (a base) into the vinegar (an acid), a reaction occurred. Salt, water, and
 _____ were produced by the reaction.

are substances that can block changes in their pH for period of time. You can find them on buffered aspirin, which keeps your ______ from dissolving the medicine too quickly. Buffers are made by combining _____ acids and bases. Buffers also are found in our body to keep our the pH of our _____ at or around 7.4.

Most science labs have pH paper, which is paper that has been soaked in a special chemical _______.
 They turn ______ in an acid and ______ in a base. By comparing the color of the pH paper to a chart, you can determine how ______ an acid or base is.

10. Constant reactions between acids and bases keep our ______ from being too acidic or too alkaline.

Brainpop Ph Scale Answer Key

Marcel A. Müller

Brainpop Ph Scale Answer Key:

The pH Scale Mary Griffin, 2018-12-15 The pH scale measures how acidic or basic a substance is ranging from 0 to 14 Readers will learn how certain substances rank on the pH scale what happens when acids and bases are mixed and how water can make a substance either acidic or basic These significant science concepts are discussed in approachable text and supported by motivating fact boxes charts images and photographs **PH Scale** ,2006 The PH Scale Marie Roesser, 2025-07-30 The pH scale measures how acidic or basic a substance is ranging from 0 to 14 This chemistry concept has real world applications For example strawberries grow best in soil with a specific pH Safe drinking water must have a pH within a certain range too Readers will learn more about pH the scale how substances are tested and how pH can change in this accessible volume a valuable addition to any elementary science collection Essential curricular information is supported by Learn More fact boxes labeled diagrams and a summarizing graphic organizer ACID-BASE CHEMISTRY NARAYAN CHANGDER, 2024-05-16 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in guiz format on our youtube channel https www youtube com smartguiziz I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging guiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests guizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more PH Scale British Standards Ph Measurement. Specification for Ph Scale British Standards Institute Staff, British Standards Institution, 1961 Institution, 1984-12-31 pH measurement pH Range statistics Accuracy Grades quality Standard solutions Phthalates Potassium organic compounds The Meaning and Calibration of the PH Scale ,1933 PH Measurements Victor Gold,1956 Ph Scale And Acid-base Titrations/standardization Postma Roberts Hollenberg, James M. Postma, Julian Roberts, J. Leland Hollenberg, 2000-01-15 **Specification for PH Scale** British Standards Institution,1950

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