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LABORATORY MANUAL SCIENCE

CLASS IX

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Short Answer Questions

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Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933 **AP* Lab Manual: Chemistry - 9th Edition**, 2014 **Practical/Laboratory Manual Chemistry Class - XI** Er. Meera Goyal, 2021-05-29 1 Basic

Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva Voce 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between Co^{2+} , H_2O , $6H^+$ and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M/10 oxalic acid solution by direct weighing method 2 To prepare M/10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce

2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 o purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

A Course in General Chemistry [with] Laboratory Manual William MACPHERSON (Professor of Chemistry, Ohio State University, and HENDERSON (William Edwards)),1913 *Joy for STEM General Chemistry Part 1 Lab Manual* C. Joy Hodnett,2021-01-08

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Table of Contents Arya Publications Lab Manual Class 9 Chemistry

1. Understanding the eBook Arya Publications Lab Manual Class 9 Chemistry
 - The Rise of Digital Reading Arya Publications Lab Manual Class 9 Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Arya Publications Lab Manual Class 9 Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Arya Publications Lab Manual Class 9 Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Arya Publications Lab Manual Class 9 Chemistry
 - Personalized Recommendations
 - Arya Publications Lab Manual Class 9 Chemistry User Reviews and Ratings
 - Arya Publications Lab Manual Class 9 Chemistry and Bestseller Lists
5. Accessing Arya Publications Lab Manual Class 9 Chemistry Free and Paid eBooks
 - Arya Publications Lab Manual Class 9 Chemistry Public Domain eBooks
 - Arya Publications Lab Manual Class 9 Chemistry eBook Subscription Services
 - Arya Publications Lab Manual Class 9 Chemistry Budget-Friendly Options
6. Navigating Arya Publications Lab Manual Class 9 Chemistry eBook Formats

- ePub, PDF, MOBI, and More
 - Arya Publications Lab Manual Class 9 Chemistry Compatibility with Devices
 - Arya Publications Lab Manual Class 9 Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Arya Publications Lab Manual Class 9 Chemistry
 - Highlighting and Note-Taking Arya Publications Lab Manual Class 9 Chemistry
 - Interactive Elements Arya Publications Lab Manual Class 9 Chemistry
8. Staying Engaged with Arya Publications Lab Manual Class 9 Chemistry
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Arya Publications Lab Manual Class 9 Chemistry
9. Balancing eBooks and Physical Books Arya Publications Lab Manual Class 9 Chemistry
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Arya Publications Lab Manual Class 9 Chemistry
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Arya Publications Lab Manual Class 9 Chemistry
- Setting Reading Goals Arya Publications Lab Manual Class 9 Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Arya Publications Lab Manual Class 9 Chemistry
- Fact-Checking eBook Content of Arya Publications Lab Manual Class 9 Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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