

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
 - Two nonmetal atoms bonded together
 - Two metal atoms bonded together
 - A metal and a nonmetal atom bonded together
 - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
16. (4 pts) How many total atoms are in this molecule: $C_6H_{12}OH$?
- 3
 - 4
 - 19
 - 20
 - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $C_5H_{10}Cl_x$, what must x be?
- 1
 - 2
 - 3
 - 4
 - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C_3H_6ClF $C_6H_5CH_3$ CS_2 $BaSO_4$
- C_3H_6ClF
 - $C_6H_5CH_3$
 - CS_2
 - $BaSO_4$
 - CS_2 and $BaSO_4$

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Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master s University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

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1. Understanding the eBook Apologia Chemistry Module 15 Study Guide
 - The Rise of Digital Reading Apologia Chemistry Module 15 Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Apologia Chemistry Module 15 Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Apologia Chemistry Module 15 Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Apologia Chemistry Module 15 Study Guide
 - Personalized Recommendations

- Apologia Chemistry Module 15 Study Guide User Reviews and Ratings
- Apologia Chemistry Module 15 Study Guide and Bestseller Lists
- 5. Accessing Apologia Chemistry Module 15 Study Guide Free and Paid eBooks
 - Apologia Chemistry Module 15 Study Guide Public Domain eBooks
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 - Apologia Chemistry Module 15 Study Guide Budget-Friendly Options
- 6. Navigating Apologia Chemistry Module 15 Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
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 - Apologia Chemistry Module 15 Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Apologia Chemistry Module 15 Study Guide
 - Highlighting and Note-Taking Apologia Chemistry Module 15 Study Guide
 - Interactive Elements Apologia Chemistry Module 15 Study Guide
- 8. Staying Engaged with Apologia Chemistry Module 15 Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Apologia Chemistry Module 15 Study Guide
- 9. Balancing eBooks and Physical Books Apologia Chemistry Module 15 Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Apologia Chemistry Module 15 Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Apologia Chemistry Module 15 Study Guide
 - Setting Reading Goals Apologia Chemistry Module 15 Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Apologia Chemistry Module 15 Study Guide
 - Fact-Checking eBook Content of Apologia Chemistry Module 15 Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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