

Introduction to Biology- EXAM #1

Enter each answer in your clicker and circle it on the paper

Choose the best answer for the following multiple choice questions.

- Organs are composed of tissues, which are composed of cells. This is an example of which characteristic of life?
 A. Living things grow and develop. B. Living things respond to stimuli.
 C. Living things maintain themselves by homeostasis. D. Living things have levels of hierarchical organization.
- The smallest unit that has all of the characteristics of life is the:
 A. cell. B. tissue. C. organ. D. organ system. E. organism.
- A fossil was once a living organism but its tissues have now been replaced by minerals and it no longer exhibits most properties of life, except for:
 A. organization B. homeostasis. C. growth & reproduction D. response to stimuli.
- Outside, the non-living world is unorganized and chaotic while inside the cell or organism there is a relatively constant environment; **maintaining this environment** is a process of:
 A. behavior. B. thermodynamics. C. homeostasis D. metabolism. E. adaptation.
- A kernel of corn contains an embryo plant that is capable of becoming a complex plant with roots, stalks, leaves and ears of corn. This potential reflects which property of life?
 A. growth & development B. response to stimuli C. adaptation D. homeostasis E. metabolism
- Which level of biological organization is **composed of tissues**?
 A. organism B. organ system C. organ D. cell E. molecules
- Which sequence correctly lists the different levels of biological organization?
 A. cells-organs-tissues-organ systems-organism B. cells-tissues-organ systems-organs-organism
 C. tissues-cells-organs-organ systems-organism D. tissues-organs-organ systems-organisms-cells
 E. cells-tissues-organs-organ systems-organisms
- Which of the following is **NOT** true about DNA?
 A. has a double helix B. bases held together by hydrogen bonds C. bases are complementary to each other
 D. has a deoxyribose sugar E. contains adenine, guanine, cytosine, and uracil
- All members of **one species** within a particular area form a:
 A. biosphere. B. ecosystem. C. population. D. nation. E. country.
- All populations** in one natural setting plus their **physical environment** constitutes a(n):
 A. biosphere. B. ecosystem. C. population. D. habitat. E. cohort.
- When researchers test a new human cancer drug using mice, the mice used constitute the:
 A. hypothesis. B. data. C. experimental design. D. model. E. control.
- Which of the following is **not** one of the six basic elements of living organisms?
 A. nitrogen B. oxygen C. chlorine D. hydrogen
- The atoms contain:
 A. neutrons only B. electrons only. C. protons, neutrons, and electrons. D. protons & neutrons.
- To measure the activity of the human brain, a **RADIOACTIVE** sugar is injected in the carotid artery and is utilized by those cells that are most active. This shows up on a PET scan and demonstrates the detection of:
 A. sulfur B. X-rays C. covalent bonds D. neutrons. E. isotopes.
- Which of the following molecules is **NOT** a compound?
 A. water or H_2O B. salt or $NaCl$ C. Oxygen or O_2 D. glucose or $C_6H_{12}O_6$
- You can hold a water droplet between two fingertips held slightly apart. This is an example of what property?
 A. cohesion of water molecules from hydrogen bonding. B. dissociation of the ionic bonding of water.
 C. you are very strong D. vibrational movement of water molecules.
- The water strider is an insect that skates across the water without sinking. The tips of its feet should be coated with molecules that repel the water or:
 A. ions B. hydrophilic C. hydrophobic D. isotopes E. acidic.
- Since pure water is neutral, it contains:
 A. no hydrogen ions B. neither hydrogen nor hydroxide ions. C. equal number of hydrogen and hydroxide ions

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Chandresh Agrawal, nandini books



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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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