

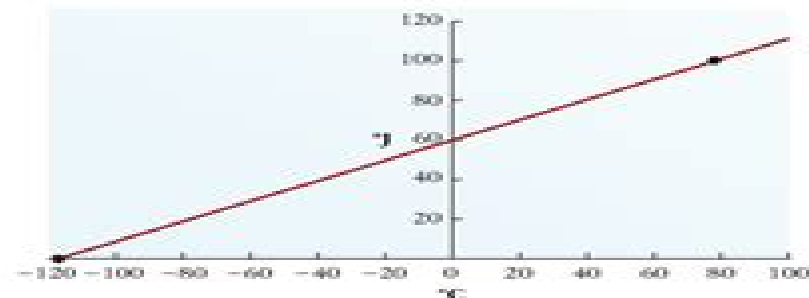
Answers to Even-Numbered and Challenge Questions and Problems

Appendix 5

Chapter 1

2. (a) element
(b) mixture
(c) compound
(d) mixture
4. (a) heterogeneous mixture
(b) heterogeneous mixture
(c) solution
6. (a) chromatography
(b) distillation or chromatography
8. (a) Cu (b) C (c) Br (d) Al
10. (a) manganese (b) gold (c) lead (d) silver
12. (a) yardstick (b) buret/pipet (c) thermometer
14. 177°C ; $4.50 \times 10^2 \text{ K}$
16. 35°C ; 95°F
18. (a) 3 (b) 4 (c) 4 (d) 1 (e) 5
20. (a) 17.25 cm (b) 169 lb
(c) $5.00 \times 10^2^{\circ}\text{C}$ (d) 198 oz
22. (a) $4.0206 \times 10^3 \text{ mL}$
(b) 1.006 g
(c) $1.001 \times 10^2^{\circ}\text{C}$
24. b and c
26. 5 is exact; 4000 is ambiguous; 17 has two and 18.5 has three
28. (a) 10.6 (b) 80.1 (c) 0.40 (d) 0.0923
(e) 1.4×10^{-22}
30. 0.04 m^3
32. (a) less than (b) greater than (c) equal
34. (a) 215.6 mL (b) 13.16 in³ (c) 0.2278 qt
36. (a) 1.15078 mi (b) 1852 m (c) 25 mph
38. (a) USD 3.09 (b) USD 43 (c) $2.1 \times 10^2 \text{ mi}$
40. $1.85 \times 10^{-3} \text{ g/mL}$
42. $3.97 \times 10^{18} \text{ L}$
44. 393.8 mg
46. 1.07 g/mL
48. 1.6 g/mL
50. 508 m
52. $3.88 \times 10^4 \text{ g}$; 85.4 lb
54. supersaturated; 5 g
56. (a) homogeneous; 4.7 g more (b) 9.1 g
58. 479 rem; $1.6 \times 10^5 \text{ mremograms}$
60. 320.0 g
62. $6.5 \times 10^{-4} \text{ in}$
64. 8.2 g/mL
66. (a) I (b) I (c) I (d) E (e) E
68. See pages 16–19
70. 3
72. (a) A (b) $\sim 23^{\circ}\text{C}$ (c) no; (see answer to (a))

73. (a)



- (b) 0.51
(c) 60
(d) $^{\circ}\text{F} = 0.512(^{\circ}\text{C}) + 32$
74. $320^{\circ}\text{F} = 160^{\circ}\text{C}$
75. 1.2 km²
76. 21.9 cm
77. $8.1 \times 10^{-3} \text{ g Pb}$
78. 6.87 g/cm^3

Chapter 2

2. See page 24
4. (a) none (b) conservation of mass
(c) none
6. Rutherford; see pages 25–26
8. (a) $86 p^{+}$ (b) 134 n
10. (a) $^{56}_{26}\text{Fe}$; $^{56}_{26}\text{Fe}$
(b) They differ in the number of neutrons. Fe-54 has 28, whereas Fe-56 has 30.
12. (a) 95 (b) 241 (c) $^{241}_{95}\text{Am}$
14. (a) A is arsenic; $33p^{+}$, 42 n , 33 e^{-}
(b) L is vanadium; $23p^{+}$, 28 n , 23 e^{-}
(c) Z is xenon; $54p^{+}$, 77 n , 54 e^{-}
16. (a) isobars; Cr-54 and Fe-54; Fe-58 and Ni-58
(b) number of protons
(c) Cr-54 and Ni-58
18. (a) < (c) < (b) < (d)
20. O-16
22. b
24. 20.2 amu
26. 64.94 amu; $^{65}_{29}\text{Cu}$
28. lightest isotope 78.6%; heaviest isotope 11.4%
30. (a) two; HCl-35 and HCl-37 (b) 36 and 38

Answers To Even Numbered Problems Seventh Edition University Physics

A Gutmann



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