BASIC ORGANIC CHEMISTRY QUESTIONS AND ANSWERS

1.	Match the names of	of compounds giv	en in column	I with their fo	rmula given	in column II
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	Column I		Column II
A	Furan	P	C_4H_5N
В	Pyrrole	Q	C_4H_4O
C	Pyridine	R	C_sH_sN
D	Tropone	S	C_7H_8O

 A-S, B-P, C-R, D-Q 	2. A-Q, B-P, C-R, D-S
A-Q, B-R, C-P, D-S	4. A-S, B-R, C-P, D-Q

1.
$$H_2C = CH_2$$
 2. $HC = C - CH = CH_2$ 3. $H_2C = C = CH_2$ 4. C_3O_2

1.
$$H_2C = CH -$$
 2. $CH_2 = C =$ 3. $CH_3 - CH =$ 4. $H - C = C -$

2. 2,4-dimethyl-3-ethylhexane

1.
$$H_2C = CH - ;vinyl$$
 2. $H_2C = CH - CH_2 - ;Allyl$
3. $HC = C - CH_2 - ;propargyl$ 4. $CH_1 - CH = CH - ;2' - propenyl$

3.
$$HC = C - CH_2 -:$$
 propargyl 4. $CH_3 - CH = CH -:$ The correct HIPAC name of the following compound is

11. Number of
$$\pi$$
 bonds in $C_2(CN)_4$ and $C_2(CN)_2$ respectively

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Book Chapter 1 interview questions and answers on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental particles Heisenberg's uncertainty principle hydrogen spectrum magnetic quantum number mass of electron metallic crystals properties Moseley law neutron properties orbital concept photons wave number Planck's quantum theory properties of cathode rays properties of positive rays quantum numbers quantum theory Rutherford model of atom shapes of orbitals spin quantum number what is spectrum x rays and atomic number The Basic Chemistry Quiz Questions PDF e Book Chapter 2 interview questions and answers on Basic chemistry atomic mass atoms molecules Avogadro s law combustion analysis empirical formula isotopes mass spectrometer molar volume molecular ions moles positive and negative ions relative abundance spectrometer and stoichiometry The Chemical Bonding Quiz Questions PDF e Book Chapter 3 interview questions and answers on Chemical bonding chemical combinations atomic radii atomic radius periodic table atomic ionic and covalent radii atoms and molecules bond formation covalent radius electron affinity electronegativity electronegativity periodic table higher ionization energies ionic radius ionization energies ionization energy periodic table Lewis concept and modern periodic table The Experimental Techniques Quiz Questions PDF e Book Chapter 4 interview questions and answers on Experimental techniques chromatography crystallization filter paper filtration filtration crucibles solvent extraction and sublimation The Gases Quiz Questions PDF e Book Chapter 5 interview questions and answers on Gas laws gas properties kinetic molecular theory of gases ideal gas constant ideal gas density liquefaction of gases absolute zero derivation applications of Daltons law Avogadro s law Boyle s law Charles law Daltons law diffusion and effusion Graham s law of diffusion ideality deviations kinetic interpretation of temperature liquids properties non ideal behavior of gases partial pressure calculations plasma state pressure units solid s properties states of matter thermometry scales and van der Waals equation The Liquids and Solids Quiz Questions PDF e Book Chapter 6 interview questions and answers on Liquid crystals types of solids classification of solids comparison in solids covalent solids properties of crystalline solids Avogadro number determination boiling point external pressure boiling points crystal lattice crystals and classification cubic close packing diamond structure dipole dipole forces dipole induced dipole forces dynamic equilibrium energy changes intermolecular attractions hexagonal close packing hydrogen bonding intermolecular forces London dispersion forces metallic crystals properties metallic solids metal s structure molecular solids phase changes energies properties of covalent crystals solid iodine structure unit cell and vapor pressure General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives

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