## Section Review 3.1

#### Part A Completion

- 1. accuracy 6. absolute value
- peecision 7, 100%
- 3. experimental value 8. scientific notation
- 4. error 9. known
- 5. accepted value 10. estimated

#### Part B True-False

11. ST 13. AT

## Part C Matching

14. b 16. a 18. d

#### Part D Questions and Problems

- 20. a. 3 c. 2
  - b. 4
- 21. a.  $4.35 \times 10^2 \, \mathrm{mL}$  e.  $2.1 \times 10^2$ 
  - b. 8.4 cm<sup>3</sup>

#### Section Review 3.2

#### Part A Completion

- 1. metric
- seven
  Celsius
- 3. meter 8. kelvin
- 4. liter 9. joule/calorie 5. weight 10. calorie/joule

## Part B True-False

11. NT 13. AT 12. NT 14. NT

### Part C Matching

- 15. k 20. l 16. e 21. d
- 17. m 22. c
- 18. g 23. h 19. a 24. i

#### Part D Questions and Problems

- 28. Volume =  $1.8 \text{ cm} \times 8.8 \text{ cm} \times 30.5 \text{ cm}$ =  $4.8 \times 10^{2} \text{ cm}^{3}$
- 29, °C = K 273 = 20 273 = -253°C
- 30. a. h; 10<sup>2</sup>
- c. m 10<sup>-1</sup>

25, 1

2566. (1)

27. b

6. gramm

- b.  $k: 10^{3}$
- $d_{\rm c} \approx 10^{-2}$

#### Section Review 3.3

#### Part A Completion

- 1. one punity:
- 6. steps
- 2. conversion factor
- 7. conversion factor

1.5 AT

- 3. remains the same
- 8. denominator
- dimensional analysis
- 10. unknown

9. umits

5. known

#### Part B True-False

- II. NT
- 13. AT
- 12. ST 14. NT

#### Part C Questions and Problems

16. a.  $125 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.125 \text{ kg}$ 

$$= 1.25 \times 10^{-1} \, \mathrm{kg}$$

- **b.**  $0.12 \text{ L} \times \frac{1000 \text{ mL}}{1 \text{ E}} = 1.2 \times 10^{2} \text{ mL}$
- 17.  $\frac{1.0 \text{ jri}}{1500 \text{ cells}} \times \frac{2.54 \text{ cris}}{1 \text{ jri}} \times \frac{1 \text{ pri}}{100 \text{ ceris}} \times \frac{10^6}{1 \text{ jri}}$ = 17  $\mu$ m/cell
- 18,  $\frac{186\,000\,\text{spdes}}{1\,\text{s}} \times \frac{1\,\text{s}}{10^6\,\mu\text{s}} \times \frac{1.61\,\text{km}}{1\,\text{gade}}$ =  $\frac{2.99 \times 10^{-1}\,\text{km}}{1}$

Land

#### Section Review 3.4

#### Part A Completion

- 1. density 3. composition
- 2. intensive

#### Part B True-False

4. ST 5. ST

#### Part C Questions and Problems

6.  $32.1 \text{ mL} \times \frac{1 \text{ cm}^3}{1 \text{ geV}} = 3.21 \text{ cm}^3$ 

Density = 
$$\frac{\text{mass}}{\text{volume}} = \frac{127 \text{ g}}{32.1 \text{ cm}^3} = 3.96 \text{ g/cm}^3$$

7.  $1.094. \times \frac{1000 \text{ cm}^3}{1 \text{ K}} = 1000 \text{ cm}^3$ 

$$1.58 \text{ kg} \times \frac{1000 \text{ g}}{\text{kg}} = 1580 \text{ g}$$

Density = 
$$\frac{\text{mass}}{\text{volume}} = \frac{1580 \text{ g}}{1000 \text{ cm}^3} = 1.58 \text{ g/cm}^3$$

picture.

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**Shayne C. Gad** 

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