Balancing Equations Worksheet II

Count the number of atoms, then use coefficients to balance each equation.

AlBr₃ + ___ K → ___ KBr + ___ Al

2) ____ FeO + ___ PdF₂ → ___ FeF₂ + ___ PdO

3) ____ P₄ + ____ Br₂ → ____ PBr₃

4) ____ LiCl + ___ Br₂ → ___ LiBr + ___ Cl₂

5) ____ PbBr₂ + ___ HCl → ___ HBr + ___ PbCl₂

6) ____ CoBr₃ + ____ CaSO₄ → ____ CaBr₂ + ____ Co₂(SO₄)₃

7) ____ Na₃P + ___ CaF₂ → ___ NaF + ___ Ca₃P₂

8) ____ Mn + ___ HI → ___ H₂ + ___ MnI₃

9) ____ Li₃PO₄ + ____ NaBr → ____ Na₃PO₄ + ____ LiBr

10) ____ CaF₂ + ____ Li₂SO₄ → ____ CaSO₄ + ____ LiF

11) ____ HBr + ___ Mg(OH)₂ → ___ MgBr₂ + ___ H₂O

12) ____ LiNO₃ + ____ CaBr₂ → ____ Ca(NO₃)₂ + ____ LiBr

13) ____ AgNO₃ + ___ Li → ___ LiNO₃ + ___ Ag

14) ____ Si(OH)₄ + ___ NaBr → ___ SiBr₄ + ___ NaOH

15) ____ NaCN + ___ CuCO₃ → ___ Na₂CO₃ + ___ Cu(CN)₂

Challenge: ____ $C_2H_6 + ___ O_2 \rightarrow ___ CO_2 + ___ H_2O$

Balancing Equations Chem Worksheet 10 2

Christian Drosten

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