

## Advanced Theory practice Test

Multiple choice (single answer) to the three main components of the question.

1. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By calculating the number of protons/neutrons
  - By identifying the number of neutrons

2. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By identifying the number of protons/neutrons
  - By identifying the number of neutrons

3. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By identifying the number of protons/neutrons
  - By identifying the number of neutrons



4. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By identifying the number of protons/neutrons
  - By identifying the number of neutrons

5. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By identifying the number of protons/neutrons
  - By identifying the number of neutrons

Figure 10



- (a) 14 protons, 14 electrons, 14 neutrons  
(b) 14 protons, 14 electrons, 14 neutrons  
(c) 14 protons, 14 electrons, 14 neutrons

6. (1.500) What is the correct order of the three components?
- By identifying the number of protons
  - By identifying the number of electrons
  - By identifying the number of protons/neutrons
  - By identifying the number of neutrons

- (a) 14 protons, 14 electrons, 14 neutrons  
(b) 14 protons, 14 electrons, 14 neutrons

# Atomic Theory Practice Test

**Marcel A. Müller**



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