Name: Date:

Balancing Chemical Equations

Balance the following chemical equations.

1.
$$\underline{1}$$
 V_2O_5 + $\underline{6}$ HCI \longrightarrow $\underline{2}$ VOCI₃ + $\underline{3}$ H₂O

2.
$$\underline{1}$$
 CH₄ + $\underline{2}$ O₂ \longrightarrow $\underline{1}$ CO₂ + $\underline{2}$ H₂O

3.
$$1$$
 Xe + 3 F₂ \longrightarrow 1 XeF₆

4.
$$2 H_2SO_4 + 1 Pb(OH)_4 \rightarrow 1 Pb(SO_4)_2 + 4 H_2O$$

7. 2 BF₃ + 3 Li₂SO₃
$$\longrightarrow$$
 1 B₂(SO₃)₃ + 6 LiF

8. 1 Be₂C + 4 H₂O
$$\longrightarrow$$
 2 Be(OH)₂ + 1 CH₄(aq)

9.
$$2 NH_4NO_3 \rightarrow 2 N_2 + 1 O_2 + 4 H_2O$$

10. 1
$$Ca_3(PO_4)_2 + 2 H_2SO_4 \rightarrow 2 CaSO_4 + 1 Ca(H_2PO_4)_2$$

Balancing Equations Sheet 61 Answer Key

Howard Anton, Chris Rorres

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