

Activity 10

Classifying reactions

Learning Objectives

- Type of chemical reactions and their mechanisms



Introduction: Chemical reactions show how **compounds** and **elements** react with one another. An element is an substance consisting of only one of a kind. Such as the elements like oxygen gas (O_2). A compound is a substance made of more than one kind of atoms. Such as water which is made up of H_2O .

Question: How are chemical reactions classified?

1. **Single:** When chemical reactions can be classified as one of four types. Using the chemical equations in the table on right, decide the following reactions in the right of reaction.

- | | |
|---|---------------------------------------|
| ☒ $2H_2 + O_2 \rightarrow 2H_2O$ (a) Redox | ☐ Redox |
| ☒ $2H_2 + O_2 \rightarrow 2H_2O$ (b) Redox | ☐ Redox |
| ☒ $2H_2 + O_2 \rightarrow 2H_2O$ (c) Redox | ☐ Redox |
| ☒ $2H_2 + O_2 \rightarrow 2H_2O$ (d) Redox | ☐ Redox |
| ☒ $2H_2 + O_2 \rightarrow 2H_2O$ (e) Redox | ☐ Redox |

2. **Double:** Reactions such as the chemical equations below. Some reactions may involve the addition of a substance to the right. Decide the reaction as a synthesis, decomposition, single replacement, or double replacement reaction.

- | | |
|---|--------------|
| (a) $2H_2 + O_2 \rightarrow 2H_2O$ Redox | Redox |
| (b) $2H_2 + O_2 \rightarrow 2H_2O$ Redox | Redox |
| (c) $2H_2 + O_2 \rightarrow 2H_2O$ Redox | Redox |
| (d) $2H_2 + O_2 \rightarrow 2H_2O$ Redox | Redox |
| (e) $2H_2 + O_2 \rightarrow 2H_2O$ Redox | Redox |

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L Manion

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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