

Rendimiento del niño en el test, observaciones y planificación de la intervención

Rendimiento del niño en el test

- Utilizando la puntuación del concepto como guía, debe anotarse un 1 o un 0 al lado de cada concepto. Cuando el niño ha respondido correctamente a un ítem del concepto, debe anotarse un 1; y cuando no ha respondido correctamente a ningún ítem del concepto, debe anotarse un 0.
- Para los niños de 3 años a 3 años 11 meses debe utilizarse la columna de la izquierda y la del medio (ítems 1/13 a 38/52).
- Para los niños de 4 años a 5 años 11 meses debe utilizarse la columna del medio y la de la derecha (ítems 53/65 a 64/76).

ÍtemsConcepto		ÍtemsConcepto		ÍtemsConcepto	
_____ 1/13	más arriba	_____ 25/39	más cerca	_____ 53/65	antes
_____ 2/14	hacia abajo	_____ 26/40	ha terminado	_____ 54/66	más lejos
_____ 3/15	vacio	_____ 27/41	más pequeño	_____ 55/67	más bajo
_____ 4/16	debajo	_____ 28/42	crucando	_____ 56/68	más corto
_____ 5/17	encima	_____ 29/43	diferente	_____ 57/69	último
_____ 6/18	falta	_____ 30/44	más largo	_____ 58/70	más abajo
_____ 7/19	al lado	_____ 31/45	delante	_____ 59/71	juntos
_____ 8/20	otra	_____ 32/46	los dos	_____ 60/72	algunas, pero no muchas
_____ 9/21	subiendo	_____ 33/47	alrededor	_____ 61/73	en medio
_____ 10/22	lleno	_____ 34/48	más alto	_____ 62/74	primer
_____ 11/23	fuera	_____ 35/49	muchos	_____ 63/75	entre
_____ 12/24	redondos	_____ 36/50	igual	_____ 64/76	menos
		_____ 37/51	más		
		_____ 38/52	más grande		

Conceptos espaciales (verde oscuro), de cantidad (verde claro), temporales (negro) y otros (gris).

Observación de los conceptos básicos por áreas de aprendizaje

Área de aprendizaje	Concepto	Comprensión receptiva	Uso expresivo	Ejemplos
Comprensión y seguimiento de las instrucciones Conceptos espaciales, de dirección y de posición, enumerales para comprender las instrucciones.	encima-debajo cruzando vacío alrededor dentro-defuera empezando-final entre empezar-justo al-entre falta primero-segundo-último hacia adelante-hacia atrás frente-espalda alto-bajo delante-detrás dentro-fuera en medio falta cerca-lejoso al lado encendido-apagado abierto-cerrado otro encima-debajo derecha-izquierda en línea salir más arriba-más abajo subiendo-bajando			

Boehm 3 Manual

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web the balanced chemical equation for this reaction is $\text{Mg} + 2\text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2$ in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

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web magnesium is already balanced on the reactant and product side balancing nitrogen by multiplying HNO_3 by 2 we get
 $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{O}_2$ magnesium nitric acid magnesium nitrate hydrogen gas oxygen and hydrogen get
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 $\text{Mg}^{2+} + \text{CO}_3^{2-} + 2\text{H}^+ \rightarrow \text{Mg}^{2+} + \text{CO}_2 + \text{H}_2\text{O}$ write the balanced equation when nitric acid reacts with calcium hydroxide to
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 reaction the mg replaces the h in the HNO_3

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 each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{NO}_2 + \text{H}_2\text{O}$ you ll need to be

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 $\text{Mg}^{2+} + 2\text{OH}^- + 2\text{H}^+ + 2\text{NO}_3^- \rightarrow \text{Mg}^{2+} + 2\text{NO}_3^- + 2\text{H}_2\text{O}$ net ionic equation $2\text{OH}^- + 2\text{H}^+ \rightarrow 2\text{H}_2\text{O}$ sulfuric acid and lithium hydroxide
 balanced total equation $\text{H}_2\text{SO}_4 + 2\text{LiOH} \rightarrow \text{Li}_2\text{SO}_4 + 2\text{H}_2\text{O}$