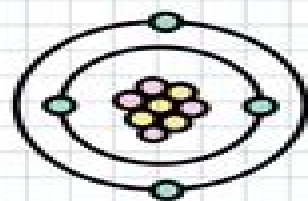


Atomic structure and the periodic table

Atomic structure



protons
neutrons
electrons

atoms have no overall charge

particle	relative mass	relative charge
proton	1	+1
neutron	1	0
electron	very small	-1

elements

elements - substances made up of atoms with the same atomic number

isotopes of an element - atoms with the same number of protons but different numbers of neutrons

relative atomic mass - the average mass number for an element

Equations, mixtures and chromatography

chemical formulas and equations

chemical formula - shows the proportion of atoms of each element in a compound

chemical equation - shows the overall change in a reaction

example:

reactants

products

word equation

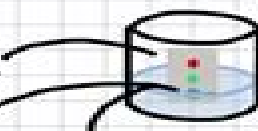
methane + oxygen → carbon dioxide + water

symbol equation



chromatogram

insoluble components stay on the base line

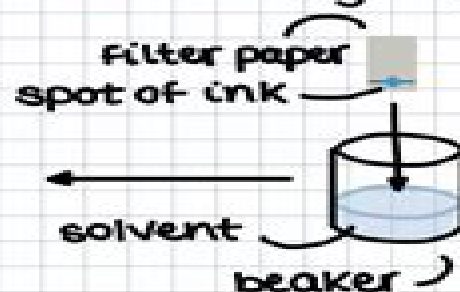


components separate as they move up the paper

submerge it



paper chromatography



filter paper

spot of ink

solvent

beaker

Atomic Structure Guided Notes Answers

Lei Huang

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