

Name _____
 Period _____

Atomic Dating Game



Partner 1 _____
 Partner 2 _____

Directions: Use your science books that you brought to school for assistance in identifying the elements. You will be using periodic information and atomic numbers to help you determine which element is the most electronegative. Use the periodic table and your science books to help you.

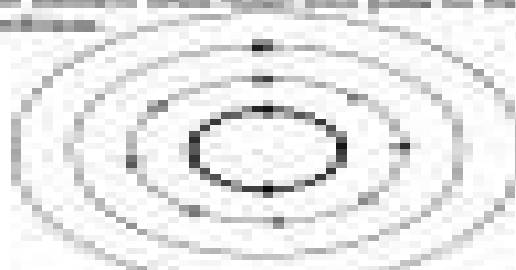
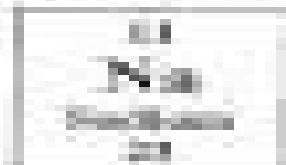
Question: What element has 9 valence electrons?

How is it an acid element?

Answer: 15 over

Chlorine group 17

Part 1: Let's compare the electronegativity of the elements. The electronegativity scale is a measure of the ability of an atom to attract electrons. The scale is based on the electronegativity of fluorine, which is the most electronegative element. The scale is based on the electronegativity of fluorine, which is the most electronegative element. The scale is based on the electronegativity of fluorine, which is the most electronegative element.

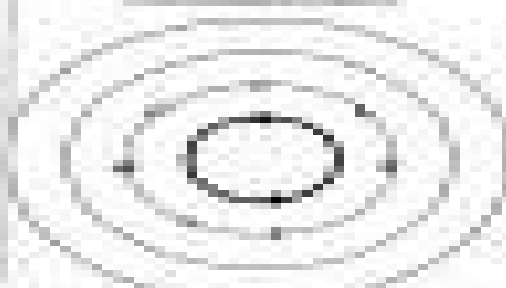
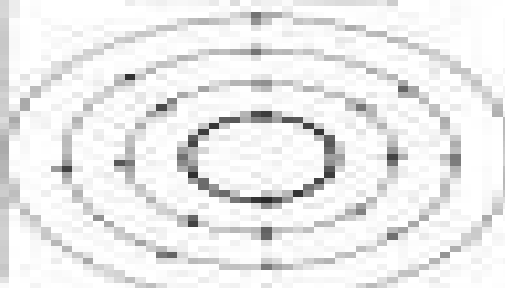
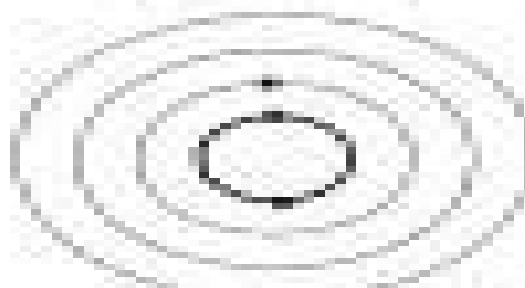
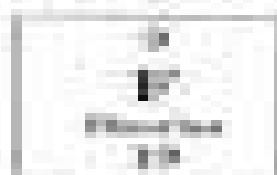


Let's compare the electronegativity of the elements.

Na: Sodium has highly reactive outer/shell, so electronegative.

Ca: Calcium is a reactive metal, so electronegative.

F: Fluorine is the most electronegative element.



Na: Sodium is a metal, so electronegative. Ca: Calcium is a metal, so electronegative. F: Fluorine is a non-metal, so electronegative.

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Siddappa N.Byrareddy



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