Chem 115: Answers to Chapter 6_Aqueous Ionic Equilibrium Practice Questions

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Practice Question 1

Calculate the pH of the buffer that results from mixing 60.0 mL of 0.250 M $HCHO_2$ and 15.0 mL of 0.500 M $NaCHO_2(K_a)$ for $HCHO_2 = 1.8 \times 10^{-4}$)

HCHO2 [A] + H20 (1)
$$\rightleftharpoons$$
 H30 [A] + CHO2 (A)

Total volume of buffer = 60.0 ml + 15.0 ml
= 75.0 ml = 0.075 L

[HCHO2] = $\frac{0.060L}{0.075L} \times 0.250M = 0.200M$

[CHO2] = $\frac{0.015L}{0.075L} \times 0.500M = 0.100M$
 $\Rightarrow pH = pKa + log [CHO2]$
[HCHO2]
= $-log (1.8 \times 10^{-4}) + log (\frac{0.100}{0.200})$
= $-(0.25527 - 4) + (-0.30103)$
= $3.74 + 7275 - 0.30103$

Aqueous Ion Equilibrium Practice

Kevin Robards, P. E. Jackson, Paul A. Haddad

Aqueous Ion Equilibrium Practice:

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Bleam, 2016-11-30 Soil and Environmental Chemistry Second Edition presents key aspects of soil chemistry in environmental science including dose responses risk characterization and practical applications of calculations using spreadsheets The book offers a holistic practical approach to the application of environmental chemistry to soil science and is designed to equip the reader with the chemistry knowledge and problem solving skills necessary to validate and interpret data This updated edition features significantly revised chapters averaging almost a 50% revision overall including some reordering of chapters All new problem sets and solutions are found at the end of each chapter and linked to a companion site that reflects advances in the field including expanded coverage of such topics as sample collection soil moisture soil carbon cycle models water chemistry simulation alkalinity and redox reactions There is also additional pedagogy including key term and real world scenarios This book is a must have reference for researchers and practitioners in environmental and soil sciences as well as intermediate and advanced students in soil science and or environmental chemistry Includes additional pedagogy such as key terms and real world scenarios Supplemented by over 100 spreadsheets to migrate readers from calculator based to spreadsheet based problem solving that are directly linked from the text Includes example problems and solutions to enhance understanding Significantly revised chapters link to a companion site that reflects advances in the field including expanded coverage of such topics as sample collection soil moisture soil carbon cycle models water chemistry simulation alkalinity and redox reactions

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