- 13. (4 pts) What is an ionic compound?
 - a. A compound that will dissolve in water
 - b. Two nonmetal atoms bonded together
 - c. Two metal atoms bonded together
 - d. A metal and a nonmetal atom bonded together
 - e. None of the above
- 14. (4 pts) What type of compound uses the naming system consisting of prefixes like mono-, di-, and tri-?
 - a. Ionic
 - b. Covalent
 - c. Both a and b
 - d. Neither a nor b
- 15. (4 pts) What type of compound uses a naming system that has -ide as the ending on the nonmetals?
 - a. Ionic
 - b. Covalent
 - c. Both a and b
 - d. Neither a nor b
- 16. (4 pts) How many total atoms are in this molecule: C₆H₁₂OH?
 - a. 3
 - b. 4
 - c. 19
 - d. 20
 - e. None of the above
- 17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is C₁H₁₀Cl₂, what must x be?
 - a. 1
 - b. 2
 - c. 3
 - d. 4
 - e. None of the above
- (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C₃H₆ClF C₆H₅CH₅ CS₂ BaSO₄
 - a. C₃H₆ClF
 - b. C₆H₃CH₃
 - e. CS₂
 - d. BaSO₄
 - e. CS2 and BaSO4

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Robert R. Redfield

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Whitaker's Book List ,1991 **Books in Series**, 1985 Vols for 1980 issued in three parts Series Authors and Titles British Books in Print ,1979 Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples **Student Study Guide/Solutions** Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Study Guide [to Accompany] General Chemistry James E. Brady, 1982 **Student Study Guide to Study Guide and Solutions Manual for** Accompany Petrucci's General Chemistry, 3rd. Ed Robert K. Wismer, 1982 Fundamentals of General, Organic, and Biological Chemistry Susan McMurry, 1992 Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons guizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor's from

Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master's University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

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