

## EXPERIMENT-2

## \* Aim

To analyse a given salt for an acid radical and basic radicals.

| I | Physical Properties<br>Exp. | Observation                                            | Inference                                                             |
|---|-----------------------------|--------------------------------------------------------|-----------------------------------------------------------------------|
| 1 | Colour                      | White                                                  | $\text{NH}_4^+$ , $\text{Pb}^{2+}$ , $\text{Al}^{3+}$ may be present. |
| 2 | Odour                       | Smell of Ammonia                                       | $\text{NH}_4^+$ may be present.                                       |
| 3 | Solubility                  | Soluble in $\text{H}_2\text{O}$                        |                                                                       |
| 4 | Strong heating test         | Smell of Ammonia evolved which gives dense white fumes | $\text{NH}_4^+$ may be present.                                       |

## II Chemical Properties.

## Test For Acid Radicals

|   | Experiment                           | Observation                                                                         | Inference.                                                                              |
|---|--------------------------------------|-------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------|
| 1 | Salt + dil $\text{H}_2\text{SO}_4$   | No reaction                                                                         | $\text{CO}_3^{2-}$ , $\text{S}^{2-}$ , $\text{SO}_3^{2-}$ & $\text{NO}_2^-$ are absent. |
| 2 | Salt + Conc. $\text{H}_2\text{SO}_4$ | Faint choking gas evolved which gives dense white fumes with $\text{NH}_4\text{OH}$ | $\text{Cl}^-$ may be present.                                                           |

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