

PHYSICAL PHARMACEUTICS

QUESTIONS:

1. Solubility of gases increases with decrease of:
a) Mass. b) Volume. c) Temperature. d) Pressure.
2. When a large amount of solute dissolves in a given amount of solvent it forms:
a) Dilute solution.
b) Saturated solution.
c) Aqueous solution.
d) Concentrated Solution.
3. Solution which can hold no more solute is called:
a) Dilute solution.
b) Saturated solution.
c) Aqueous solution.
d) Concentrated Solution.
4. Which is not a colligative property:
a) Raoult's law.
b) Depression in freezing point.
c) Osmotic pressure.
d) Melting point.
5. Azeotropic mixtures are:
a) Constant temperature boiling mixtures.
b) Those which boils at different temperatures.
c) Mixture of two solids.
d) None of above.
6. The pH of pharmaceutical buffer system can be calculated by:
a) pH partition theory.
b) Noyes Whitney law.
c) Henderson - Hasselbalch equation.
d) Michaelis Menten equation.
7. Unit of surface tension is:
a) Dyne/ meter
b) Dyne/cm
c) Cg/cm
d) None of the above.
8. Two solutions are said to be isotonic if they exert same.....
a) Viscosity. c) Osmotic pressure.
b) Surface tension. d) None of the above.
9. Dissolution is affected by:
a) Surface area.
b) Viscosity.
c) Temperature.
d) All of the above.

Basic Physical Pharmacy Questions With Answer

Manan Shroff



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Basic Physical Pharmacy Joseph K. H. Ma, Boka Hadzija, 2013 Basic Physical Pharmacy provides a thorough yet accessible overview of the principles of physical pharmacy and their application in drug formulation and administration This definitive guide to physical pharmacy covers all types of pharmaceuticals from traditional forms and dosages to nanotechnology based novel dosage design Authored by two nationally recognized pharmaceutical scientists and active pharmacy faculty Basic Physical Pharmacy is clearly organized into four sections Physical Pharmacy in Solutions Solid Dosage Forms Polyphasic Systems and Drug Delivery and Novel Drug Delivery Systems Students can build upon their chemistry education to learn the physicochemical properties of drugs and their therapeutic effects on the body With a highly accessible approach Basic Physical Pharmacy will help students comprehend and apply the principles of physical pharmacy in clinical practice Covers major drug products and delivery systems Features current trends in pharmaceutical research and development including nanotechnology based dosage design Includes many examples of useful equations and formulation methods Contains over 200 illustrations photos and tables Topics Include Solutions Ionization of Drugs in Solutions Buffers and Buffered Solutions Drug Solubility Diffusion and Dissolution Distribution Phenomena Complexation and Protein Binding Interfacial Phenomena Rheology Colloids Suspensions and Emulsions Semisolid Dosage Forms Dermatologicals Suppositories Powders Capsules Tablets Aerosols Sterile Dosage Forms Ophthalmic Formulations Radiopharmaceuticals Modified Release Drug Delivery Systems Biotechnology Products Drug Product Stability Each new print textbook includes an access code for the online Companion Website Ebooks do not include access to the Companion Website Access to the Companion Website may also be purchased separately under the RESOURCES tab FOR STUDENTS Student Companion Website includes Cross Words Flash Cards Interactive Glossary Matching Questions Instructor Resources Answers to End of Chapter Questions Image Bank Power Point Presentations Test Bank

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